

**UNIVERSIDADE FEDERAL DE UBERLÂNDIA  
INSTITUTO DE BIOTECNOLOGIA  
PÓS-GRADUAÇÃO EM BIOTECNOLOGIA**

**TALLITA STÉFANNE E SILVA**

**DESENVOLVIMENTO DE UM IMUNOSENSOR IMPEDIMÉTRICO PARA A  
DETECÇÃO DO VÍRUS SINCICIAL RESPIRATÓRIO**

**PATOS DE MINAS – MG  
JUNHO DE 2023**

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Dissertação de Mestrado apresentada ao  
Programa de Pós-graduação em Biotecnologia  
como requisito parcial para a obtenção do título  
de Mestre em Biotecnologia.

**Orientador: Prof. Dr. Diego Leoni Franco**

**PATOS DE MINAS – MG**

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Aprovado em \_\_\_/\_\_\_/\_\_\_

**BANCA EXAMINADORA**

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Prof. Dr. Diego Leoni Franco (Orientador)

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Prof. Dr. Guilherme Ramos Oliveira e Freitas

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Prof. Dr. Lucas Franco Ferreira

**PATOS DE MINAS – MG**

**2023**

*Dedico esta dissertação a todos os meus professores: do ensino fundamental até o ensino superior.*

Remember the moment  
You left me alone and  
Broke every promise you ever made  
I was an ocean lost in the open  
Nothing could take the pain away

So you can throw me to the wolves  
Tomorrow I will come back  
Leader of the whole pack  
Beat me black and blue  
Every wound will shape me  
Every scar will build my throne

The sticks and the stones that  
You used to throw have  
Built me an empire, so don't even try  
To cry me a river, 'cause I forgive you  
You are the reason I still fight

So you can throw me to the wolves  
Tomorrow I will come back  
Leader of the whole pack  
Beat me black and blue  
Every wound will shape me  
Every scar will build my throne

**THRONE**  
**BRING ME THE HORIZON**



*“Somewhere, something incredible is waiting to be known.”*

– Carl Sagan

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## RESUMO

O vírus sincicial respiratório (RSV) é um importante patógeno respiratório que acomete principalmente crianças menores de cinco anos. Apesar de na maioria dos casos de infecção em pessoas adultas e imunocompetentes causar quadros leves sem complicações, em crianças e idosos é frequente o aparecimento de bronquiolite, bronquite e pneumonia. As formas de detecção do RSV incluem RT-qPCR, ensaio imunocromatográfico e ensaio de imunofluorescência indireto que apresentam desvantagens relacionadas ao tempo, preço e sensibilidade que podem ser contornadas pelo desenvolvimento de um imunossensor eletroquímico, um dispositivo de baixo custo, rápido, específico e de fácil manuseio. A ligação covalente é uma metodologia efetiva para a imobilização do anticorpo (Ab), de modo que a modificação eletroquímica do eletrodo de trabalho de carbono grafite de lapiseira (PGE), com polímeros condutores ou não, que apresentem grupos funcionais, são interessantes para esse procedimento. O ácido 3-aminofenilacético (3APA) apresenta as características desejadas e foi o monômero escolhido, com intuito de apresentar grupos carboxilas disponíveis após a formação polimérica. As técnicas de voltametria cíclica (CV) e de espectroscopia de impedância eletroquímica (EIS) foram usadas para a eletropolimerização e caracterização do material formado. Os CVs demonstraram que houve formação de poli(3APA) e 30 ciclos foram estabelecidos por apresentar a melhor resposta para o sistema. Estudos eletroquímicos e de pH em CV possibilitaram determinar o número de elétrons e a razão próton/elétron para apresentar a proposta do mecanismo de eletropolimerização do 3APA. O imunossensor foi desenvolvido sobre esta matriz com a ativação dos grupos carboxílicos do poli(3APA) em solução de 1-etil-3-(3-dimetilaminopropil)carbodiimida (EDC) / N-hidroxisuccinimida (NHS) seguido da reação de substituição nucleofílica com os grupos amino presentes no Ab. Após a imobilização de Ab, o PGE/poli(3APA) teve sua superfície bloqueada por glicina (Gly) para impedir interações inespecíficas, seguido da imobilização do antígeno (Ag). As detecções ocorreram por EIS e os dados foram avaliados pelos circuitos equivalentes. Houve o aumento nos valores de resistência à transferência de carga ( $R_{tc}$ ) conforme o PGE foi modificado, na seguinte ordem: PGE < PGE/poli(3APA) < PGE/poli(3APA)/Ab < PGE/poli(3APA)/Ab/Ag. O sistema foi otimizado buscando as melhores condições para detecção. Foi usada a variação de resposta ( $\Delta R_{tc}$ ) entre o sistema completo contendo Ab/Ag e o sistema contendo somente Ab ( $n = 3$ ). Para o preparo do imunossensor foram escolhidos 100 ng de Ab, e o tempo de imobilização de Ab e Gly foram de, respectivamente, 3 h e 50 min. O tempo de resposta do sistema, com a imobilização de Ag foi de 30 min. Com a curva de calibração do sistema otimizado, o LOD calculado foi de 27,65 PFU.mL<sup>-1</sup> e o LOQ de 92,15 PFU.mL<sup>-1</sup>, demonstrando que o imunossensor proposto é uma alternativa viável para a detecção de RSV.

**Palavras-chave:** Vírus sincicial respiratório; Imunossensor eletroquímico; Ácido 3-aminofenilacético; Eletropolimerização; Voltametria cíclica; Espectroscopia de impedância eletroquímica.

## ABSTRACT

Respiratory syncytial virus (RSV) is an important respiratory pathogen that mainly affects children under the age of five. While RSV infections in healthy, immunocompetent adults usually result in mild uncomplicated cases, infants and the elderly often experience bronchiolitis, bronchitis, and pneumonia. Current detection methods for RSV include RT-qPCR, immunochromatography, and indirect immunofluorescence assays, but they have disadvantages related to time, cost, and sensitivity. These limitations can be overcome by developing an electrochemical immunosensor, a low-cost, rapid, specific, and user-friendly device. Covalent binding is an effective methodology for antibody (Ab) immobilization, and the electrochemical modification of a graphite pencil lead working electrode (PGE) with conductive or non-conductive polymers that possess functional groups is of interest for this procedure. 3-aminophenylacetic acid (3APA) exhibits the desired characteristics and was chosen as the monomer to provide available carboxyl groups after polymerization. Cyclic voltammetry (CV) and electrochemical impedance spectroscopy (EIS) techniques were employed for electropolymerization and characterization of the formed material. The CV results demonstrated the formation of poly(3APA), with 30 cycles established as the optimal condition due to the best response observed in the system. Electrochemical and pH studies in CV allowed for the determination of the number of electrons and the proton-to-electron ratio, providing insight into the proposed mechanism of 3APA electropolymerization. The immunosensor was developed on this matrix by activating the carboxylic groups of poly(3APA) in a solution of 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide (EDC) / N-hydroxysuccinimide (NHS), followed by nucleophilic substitution reaction with the amino groups present in the antibody (Ab). After Ab immobilization, the PGE/poly(3APA) surface was blocked by glycine (Gly) to prevent non-specific interactions, followed by antigen (Ag) immobilization. Detection was performed using EIS, and the data were evaluated using equivalent circuits. The values of charge transfer resistance ( $R_{tc}$ ) increased as the PGE was modified in the following order: PGE < PGE/poly(3APA) < PGE/poly(3APA)/Ab < PGE/poly(3APA)/Ab/Ag. The system was optimized to achieve the best detection conditions. The response variation ( $\Delta R_{tc}$ ) between the complete system containing Ab/Ag and the system containing only Ab ( $n = 3$ ) was utilized. For the preparation of the immunosensor, 100 ng of Ab was chosen, and the immobilization times for Ab and Gly were, respectively, 3 h and 50 min. The response time of the system, with Ag immobilization, was 30 min. Using the calibration curve of the optimized system, the calculated limit of detection (LOD) was 27.65 PFU.mL<sup>-1</sup>, and the limit of quantification (LOQ) was 92.15 PFU.mL<sup>-1</sup>, demonstrating that the proposed immunosensor is a viable alternative for the detection of RSV.

**Keywords:** Respiratory syncytial virus; Electrochemical immunosensor; 3-aminophenylacetic acid; Electropolymerization; Cyclic voltammetry; Electrochemical impedance spectroscopy.

## LISTA DE ABREVIATURAS E SIGLAS

3APA	Ácido 3-aminofenilacético
Ab	Anticorpo
Ag	Antígeno
cDNA	DNA complementar
CV	Voltametria cíclica
DNA	Ácido desoxirribonucleico
EDC	1-etil-3-(3-dimetilaminopropil)carbodiimida
EIS	Espectroscopia de impedância eletroquímica
F	Glicoproteína de fusão
Gly	Glicina
IgG	Imunoglobulina G
IRA	Infecção respiratória aguda
IRATRI	Infecção respiratória aguda no trato respiratório inferior
IRATRS	Infecção respiratória aguda no trato respiratório superior
L	RNA polimerase dependente de RNA viral
LOD	Limite de detecção
LOQ	Limite de quantificação
M	Matriz
mRNA	RNA mensageiro
N	Nucleoproteína
NHS	N-hidroxisuccinimida
P	Fosfoproteína
PCR	<i>Polymerase chain reaction</i>
qPCR	<i>Real time</i> PCR
RNA	Ácido ribonucleico
RSV	Vírus sincicial respiratório
RT-qPCR	<i>Reverse transcription</i> qPCR
Rtc	Resistência a transferência de carga
SH	Gliproteína hidrofóbica pequena

## LISTA DE SÍMBOLOS E UNIDADES

$\beta$	Beta
$\pi$	Pi
$\sigma$	Sigma
%	Por cento
g	Gramma
h	Hora
min	Minutos
mL	Mililitro
ng	Nanogramas
PFU	Unidade formadora de placa

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## CAPÍTULO 1: REFERENCIAL TEÓRICO

### 1 INTRODUÇÃO

O vírus sincicial respiratório (RSV) é um patógeno responsável por infecções nas vias respiratórias, majoritariamente em crianças menores de cinco anos. Em casos graves, há infecção respiratória aguda no trato respiratório inferior (IRATRI), podendo ocasionar em pneumonia e bronquiolite, doenças presentes entre as principais causas de morbidade e mortalidade em pacientes com menos de 5 anos (BORCHERS et al., 2013; LIU et al., 2016; SHI et al., 2017). No geral, o diagnóstico é feito por avaliação clínica, mas devido ao elevado número de microrganismos responsáveis por ocasionar os mesmos quadros do RSV, exames de diagnóstico laboratorial podem ser necessários (JHA et al., 2016; RALSTON et al., 2017). O ensaio de imunofluorescência indireto é a principal técnica para detecção do vírus, com boa sensibilidade e baixo custo, mas apresenta menor agilidade em sua execução. Outra alternativa é a *reverse transcription real time quantitative polymerase chain reaction* (RT-qPCR), vantajosa devido à alta sensibilidade. Porém, é uma técnica cara em relação a equipamentos e reagentes e necessita de profissionais qualificados. Também existem testes imunocromatográficos para detecção de antígenos, conhecidos popularmente como testes rápidos, que são baratos e específicos, mas com sensibilidade de apenas 75.3% durante surtos do vírus (BRUNING et al., 2017; JAIN; SCHWEITZER; JUSTICE, 2022). Desse modo, é necessário o desenvolvimento de uma forma de detecção que una as vantagens dos testes já existentes, ou seja, uma técnica rápida, sensível, específica e de baixo custo.

Biossensores são capazes de detectar analitos específicos através da conversão de um sinal bioquímico, proveniente de um componente biológico imobilizado, para um sinal detectável por meio de um transdutor (SCHULTZ; TAYLOR, 1996), que pode ser óptico (VÉDRINE et al., 2003), eletroquímico (STÉFANNE E SILVA et al., 2020), ou piezoelétrico (ZHOU et al., 2002), entre os mais comuns. Além disso, podem ser imobilizadas diferentes tipos de moléculas biológicas, como enzimas (STÉFANNE E SILVA et al., 2020), anticorpos (VENTURINI ULIANA; YAMANAKA, 2019), ácidos nucleicos (DA FONSECA ALVES et al., 2019) e até mesmo células inteiras (VÉDRINE et al., 2003). A aplicação de biossensores se estende por uma variedade de áreas (MEHROTRA, 2016), como a ambiental (STÉFANNE E SILVA et al., 2020; VÉDRINE et al., 2003; VENTURINI ULIANA; YAMANAKA, 2019), a alimentícia (SOARES et al., 2019) e a da saúde (DA FONSECA ALVES et al., 2019; RAMANATHAN; JÖNSSON; DANIELSSON, 2001; ZHOU et al., 2002).

Os anticorpos, componentes biológicos que atuam no sistema imunológico (ABBAS; LITCHTMAN; PILLAI, 2019a), podem ser imobilizados em um transdutor para o desenvolvimento de imunossensores. A ligação antígeno-anticorpo é específica e a sensibilidade do dispositivo dependerá da eficiência da imobilização, da estabilidade da biomolécula imobilizada da afinidade entre essa reação, em que ligações irreversíveis são vantajosas, pois garantem alta sensibilidade (GIZELI; LOWE, 1996; HOCK, 1997). Sendo assim, o desenvolvimento de um imunossensor é interessante para diagnóstico de RSV.

Entre os transdutores, o eletroquímico apresenta como vantagens a miniaturização e a sensibilidade. Nesse tipo de dispositivo, um eletrodo de trabalho faz a conversão do sinal bioquímico em um sinal elétrico quantificável, possível devido ao fluxo de elétrons presentes na solução a ser analisada, em que o valor adquirido é proporcional à concentração do analito (ABDULBARI; BASHEER, 2017; WANG, 1996, 2000). Eletrodos a base de carbono são comumente usados no desenvolvimento de biossensores. Eles podem ser carbono vítreo (RAMANATHAN; JÖNSSON; DANIELSSON, 2001), pasta de carbono (ZHANG et al., 2007), nanotubos de carbono (WANG et al., 2003), carbono de grafite de lapiseira (STÉFANNE E SILVA et al., 2020), entre outros. O eletrodo de carbono grafite de lapiseira (PGE) tem mostrado resultados tão satisfatórios quanto os demais eletrodos a base de carbono, além de ser uma alternativa de baixo custo, de fácil acesso e de fácil manuseio (ALVES et al., 2017; DA FONSECA ALVES et al., 2019; SOARES et al., 2019; STÉFANNE E SILVA et al., 2020; TORRINHA et al., 2018).

Para o desenvolvimento de biossensor, é necessária uma forma eficiente de imobilização do componente biológico sobre o transdutor. Os tipos de imobilização mais comuns são ligação covalente, aprisionamento, *cross-linking* e adsorção física (TAYLOR, 1996). No método de ligação covalente, o eletrodo de trabalho pode ter sua superfície modificada com um filme polimérico, condutor ou não, geralmente formado por voltametria cíclica (CV), técnica eletroquímica que oferece informações qualitativas sobre a modificação. O principal objetivo dessa modificação é obter um polímero com grupos funcionais disponíveis para a ligação com o componente biológico (BATFLETT, 1996; PACHECO et al., 2013). Monômeros derivados da anilina, como o ácido 3-aminofenilacético (3APA), são bons candidatos para eletropolimerização na superfície do eletrodo devido ao possível caráter condutor desses materiais, o que melhora a resposta de detecção do analito (TAHIR; ALOCILJA; GROOMS, 2005).

Desta forma, este trabalho possui dois grandes objetivos: a modificação de PGEs com polímeros derivados do 3APA, seguido da caracterização eletroquímica do material formado e

utilização desta plataforma modificada para o desenvolvimento de um imunossensor impedimétrico para o RSV. Até onde temos conhecimento, não há na literatura um biossensor eletroquímico baseado em PGEs modificados com polímero derivado de 3APA para esta aplicação.

## **2 REFERENCIAL TEÓRICO**

### **2.1 Infecção respiratória aguda (IRA)**

Infecções respiratórias agudas (IRAs) são caracterizadas pela obstrução da passagem de ar no sistema respiratório e é uma das principais causas de hospitalização de crianças em todo o mundo (BERMAN, 1991). Elas podem ser classificadas como infecção respiratória aguda no trato respiratório superior (IRATRS) e infecção respiratória aguda no trato respiratório inferior (IRATRI). A IRATRS atinge desde as vias aéreas das narinas até as cordas vocais na laringe, e é a doença mais frequente em humanos e a mais comum na infância, já que crianças com até 5 anos desenvolvem, em média, de 6 a 8 infecções por ano (MONTO, 1994; SIMOES et al., 2006). Apesar de não gerar casos graves, IRATRS podem causar epidemias generalizadas que permitem a circulação constante de patógenos (YEN et al., 2019).

A IRATRI atinge a traqueia, brônquios, bronquíolos ou alvéolos, ocasionando em quadros de bronquiolite, bronquite e pneumonia (BASILIO et al., 2017; SIMOES et al., 2006). Por desenvolver casos graves, as IRATRIs requerem bastante atenção, principalmente por serem a maior causa infecciosa de morte em infectados com até 5 anos de idade (YEN et al., 2019). Dados da OMS de 2000 a 2003 demonstram que 19% das mortes de crianças dessa faixa etária foram causadas por pneumonia (BRYCE et al., 2005). Os três principais causadores da doença são as bactérias *Haemophilus influenzae* tipo b e *Streptococcus pneumoniae* (imunopreveníveis) e o vírus sincicial respiratório (ainda sem imunoprevenção para essa faixa etária) (HARRIS, 2023; NAIR et al., 2010).

### **2.2 Vírus sincicial respiratório (RSV)**

O RSV, pertencente ao gênero *Orthopneumovirus* e família *Pneumoviridae* (ICTV, 2021), foi isolado pela primeira vez de chimpanzés em 1956 (MORRIS; BLOUNT; SAVAGE, 1956) e posteriormente, em 1957, de bebês (CHANOCK; ROIZMAN; MYERS, 1957). O vírus é um patógeno humano comum, caracterizado por não desenvolver imunidade a longo prazo,

acarretando em reinfecções ao longo da vida. O primeiro contágio, em 90% dos casos, acontece até os 2 anos de idade (JAIN; SCHWEITZER; JUSTICE, 2022). Comumente, seu quadro clínico é caracterizado por IRATRS, sem grandes complicações, mas, em menores proporções, pode causar IRATRI em crianças, idosos e adultos imunocomprometidos. Devido a isso, o vírus se tornou um patógeno importante, podendo gerar surtos em todo o mundo, principalmente em países subdesenvolvidos (JHA et al., 2016; SHI et al., 2017). O RSV se apresenta em dois subgrupos antigênicos (A e B) que normalmente circulam juntos, mas com predominância de um subtipo sobre o outro. (MUFSON et al., 1985).

### ***2.2.1 Epidemiologia***

Aproximadamente 30 milhões de casos de RSV acontecem anualmente, acarretando em mais de 100 mil mortes de crianças. Os países subdesenvolvidos apresentam alta taxa de mortalidade pelo vírus, devido ao diagnóstico tardio, correspondendo a 90% dos óbitos pela infecção viral (PATH, 2017). Em pacientes com menos de 5 anos, é comum que o patógeno cause bronquiolite ou pneumonia devido a IRATRI (JAIN; SCHWEITZER; JUSTICE, 2022). Dentre as internações e as mortes hospitalares causadas pelo vírus, 45% correspondem a crianças com menos de 6 meses (SHI et al., 2017).

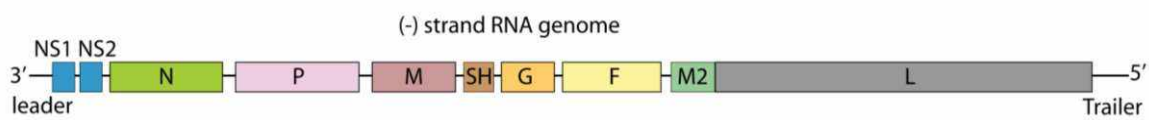
No Brasil, estudos realizados em um hospital de São Paulo – São Paulo entre 2009 e 2013, demonstraram que 14% das crianças infectadas com RSV apresentaram quadro mais grave e precisaram ser internadas. A idade média desse grupo era de 1,6 anos, enquanto crianças infectadas, mas sem complicações tinham idade média de 3,1 anos, demonstrando que pacientes mais novos são mais vulneráveis à infecção (CRUZ et al., 2021). Um estudo realizado em pacientes de até 5 anos em um hospital de Uberlândia – Minas Gerais entre 2010 e 2015 demonstrou que a detecção de RSV em crianças hospitalizadas ocorreu em uma frequência 2,3 vezes maior do que outros vírus respiratórios (rhinovirus, influenzavirus, parainfluenzavirus, adenovirus e metapneumovirus), demonstrando uma maior gravidade do patógeno e reafirmando a sua importância (COSTA et al., 2022)

### ***2.2.2 Estrutura e replicação***

O RSV é um vírus envelopado de RNA fita simples, não segmentado, de sentido negativo pertencente a classe V do sistema de classificação de Baltimore. Seu genoma (**Figura 1**), com duas janelas de leitura aberta, é composto por 10 genes que codificam 11 proteínas

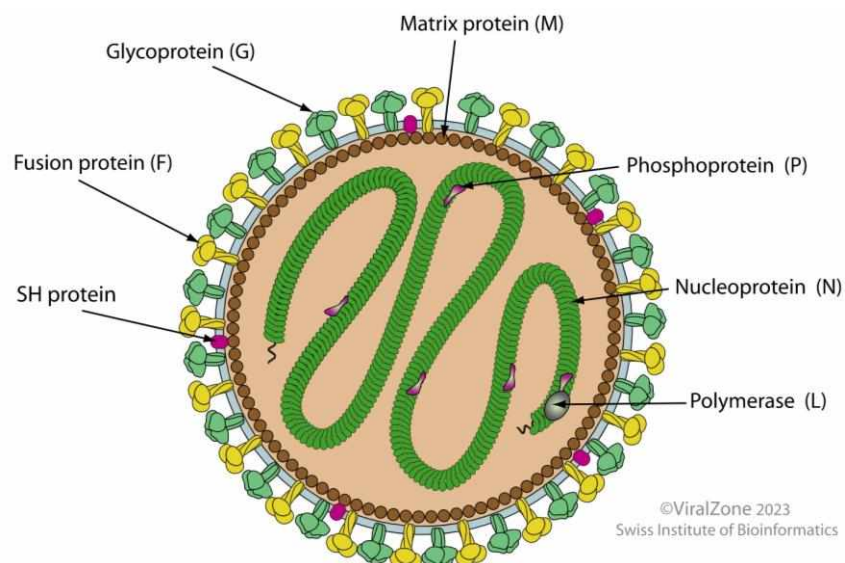
(BORCHERS et al., 2013). As partículas virais (**Figura 2**) são envoltas por um envelope lipídico que apresenta 3 glicoproteínas de superfície: a glicoproteína G, que garante a ligação do vírus à superfície da célula hospedeira; a glicoproteína de fusão (F), responsável pela fusão das membranas virais e celulares; e a glicoproteína hidrofóbica pequena (SH), presente em menor quantidade nas partículas virais, mas que é expressa em abundância na superfície das células infectadas. O nucleocapsídeo contém outras 6 proteínas estruturais: a RNA polimerase dependente de RNA viral (L), relacionada com várias atividades enzimáticas necessárias para a replicação; a nucleoproteína (N), que se liga fortemente ao RNA viral e evita sua degradação; a fosfoproteína (P), que impede a ligação entre a proteína N e os RNAs celulares; M2-1 e M2-2, codificadas a partir do gene M2, com duas janelas de leitura aberta e agem como reguladoras da transcrição e replicação viral; e a matriz (M), envolvida na coordenação da montagem das proteínas do envelope e do nucleocapsídeo, além de auxiliar na formação de novas partículas virais imaturas através da membrana celular hospedeira. Ademais, existem duas proteínas não estruturais: NS1 e NS2 (JHA et al., 2016; MELERO, 2007).

**Figura 1** – Estrutura molecular do vírus sincicial respiratório humano. NS: proteínas não estruturais (NS1 e NS2), N: nucleoproteína, P: fosfoproteína, M: proteína de matriz, SH: glicoproteína hidrofóbica pequena, G: glicoproteína, F: proteína de fusão, M2: proteína de matriz, L: RNA polimerase dependente de RNA viral.



Fonte: ViralZone, 2023.

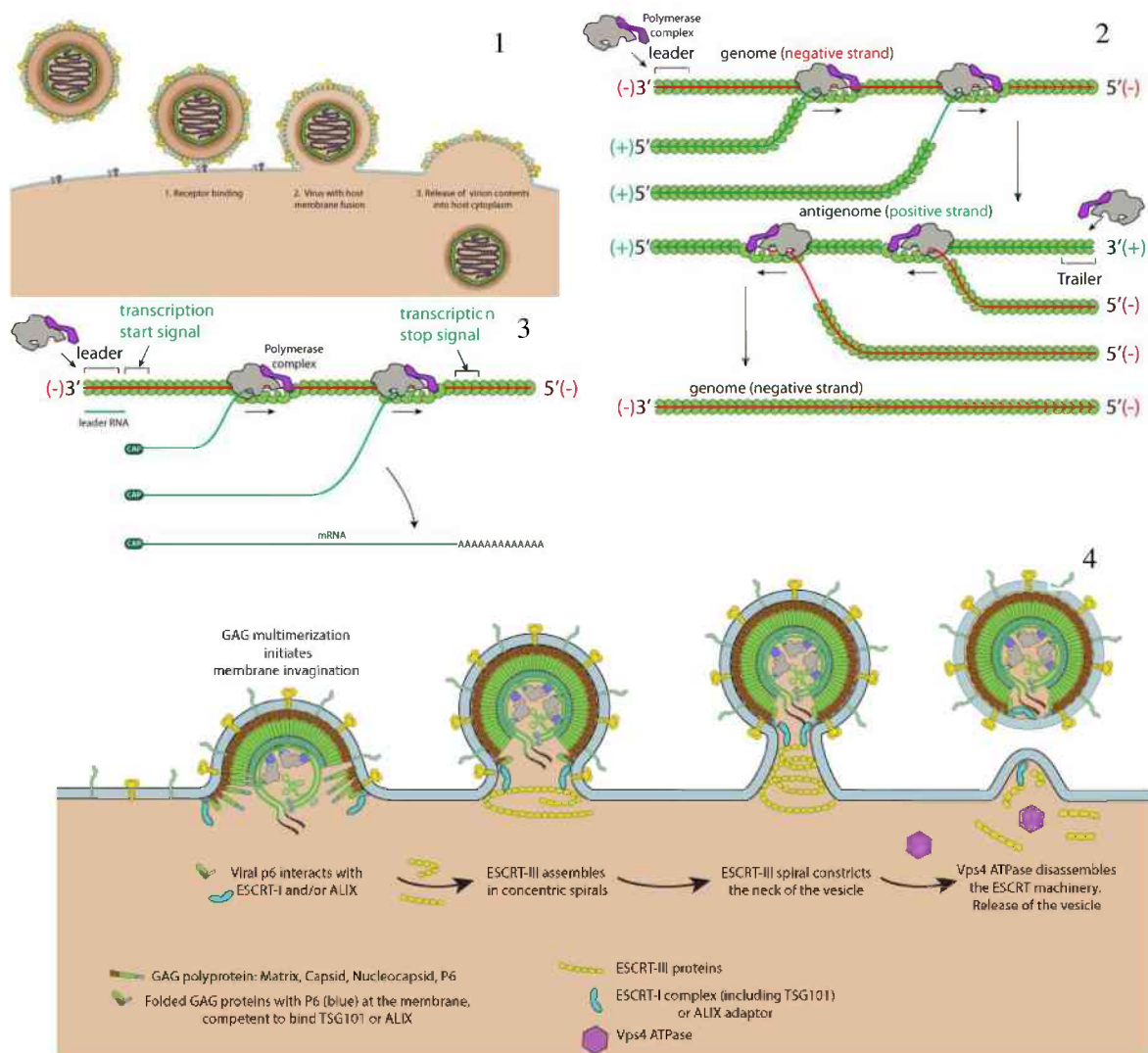
**Figura 2** – Estrutura do vírion do vírus sincicial respiratório humano.



Fonte: ViralZone, 2023.

Quando o vírus entra em contato com a célula hospedeira, ocorre a ligação inicial entre a partícula viral e os receptores presentes na superfície celular pela glicoproteína G. Posteriormente, a proteína F é ativada para proporcionar a fusão, possibilitando a entrada do nucleocapsídeo viral no citoplasma da célula hospedeira. No meio intracelular, o RSV ativa a transcrição de seus genes de infecção por meio de um grupo de mRNAs que condiciona a síntese proteica aos ribossomos celulares. A replicação ocorre um pouco depois, gerando uma cópia completa de seu antígenoma (RNA viral de sentido positivo) que formará um complexo com a proteína N e servirá como molde para a síntese do genoma das novas partículas. As proteínas recém traduzidas migram para perto da membrana celular, onde ocorre a montagem dos novos vírus que, em seguida, serão liberados por brotamento (**Figura 3**) (COLLINS; FEARNES; GRAHAM, 2013; MELERO, 2007).

**Figura 3** – Ciclo infeccioso do vírus sincicial respiratório humano. Fusão (1), transcrição sequencial (2), replicação (3) e liberação para o meio extracelular (4).



Fonte: ViralZone, 2023.

### **2.2.3 Fisiopatologia**

A entrada do vírus no organismo do hospedeiro ocorre na mucosa do trato respiratório superior ou conjuntiva através do contato direto ou por gotículas contaminadas. O tempo de incubação é de 2-8 dias, variando de acordo com aspectos do hospedeiro, como idade ou quantidade de infecções anteriores. Em seguida, o vírus se espalha rapidamente, até chegar ao epitélio, onde seu crescimento é favorecido (JAIN; SCHWEITZER; JUSTICE, 2022; JHA et al., 2016).

Os mecanismos de imunidade inata são os primeiros que agem na defesa do hospedeiro, gerando uma resposta rápida, nas primeiras horas ou nos primeiros dias de infecção. Sendo assim, utiliza mecanismos já existentes no corpo, enquanto a resposta imune adaptativa está em desenvolvimento (ABBAS; LITCHTMAN; PILLAI, 2019b). Na infecção pelo RSV, a imunidade inata desempenha um importante papel na patogênese. As células epiteliais presentes nas vias aéreas, bem como os macrófagos alveolares e as células dendríticas, são responsáveis por ativar a imunidade inata e desenvolver a imunidade adaptativa contra o vírus. Essa etapa gera resposta inflamatória através da indução de citocinas e quimiocinas (TAYLOR, 2006).

A imunidade adaptativa não tem efeito imediato no organismo, pois é desenvolvida após o contato com o agente infeccioso. Ao contrário da imunidade inata, ela é específica e especializada, além de fortalecer seus mecanismos de defesa a cada novo contágio (ABBAS; LITCHTMAN; PILLAI, 2019b). A infecção por RSV não garante total proteção contra reinfecções, mas concede acúmulo de resistência à cada nova reinfecção do trato respiratório inferior, demonstrando que a imunidade é mais duradoura na infecção pulmonar do que no trato respiratório superior (TAYLOR, 2006). A resposta imunológica adaptativa ao vírus inclui a ativação de linfócitos humoral e citotóxico que, juntamente com a atividade citotóxica viral e a atividade citotóxica do hospedeiro, causam a necrose das células epiteliais do trato respiratório, ocasionando em uma pequena obstrução das vias aéreas por muco, e em casos mais graves, na obstrução dos alvéolos (JAIN; SCHWEITZER; JUSTICE, 2022).

### **2.2.4 Diagnóstico**

O diagnóstico na maioria das infecções de RSV, sem complicações, é feito através da avaliação clínica dos sintomas do paciente. Porém, casos de IRATRI requerem maior cuidado devido a existência de bactérias causadoras da mesma infecção. Nessas circunstâncias, é preciso

confirmar que o RSV é o patógeno responsável pela infecção, afim de evitar a prescrição errônea de antibióticos (RALSTON et al., 2017). A detecção pode ser realizada quando o patógeno chega ao epitélio do hospedeiro, favorecendo o crescimento do vírus e aparição de sintomas no infectado (JAIN; SCHWEITZER; JUSTICE, 2022).

O ensaio de imunofluorescência indireto é o método mais comum no diagnóstico de RSV (TANG; LOH, 2014). Nele são usados dois anticorpos: o anticorpo primário, que garante a especificidade da reação; e um anti-anticorpo conjugado com fluorocromo que revela a reação (ABBAS; LICHTMAN; PILLAI, 2019c). A técnica possui boa sensibilidade, é relativamente rápida (1-3 h) e tem baixo custo, porém seu rendimento é lento e necessita de mão de obra qualificada na operação, principalmente para avaliação dos resultados (MADELEY; PEIRIS, 2002).

Os ensaios imunocromatográficos, comumente conhecidos como testes rápidos, economizam tempo, com resultados prontos em 10-30 minutos, e são simples de serem manuseados, possibilitando o diagnóstico no próprio local e hora do atendimento ao paciente (JAIN; SCHWEITZER; JUSTICE, 2022). Estudos demonstram que os testes de primeira geração, baseados na mudança macroscópica de cor em contato com o antígeno, possuem sensibilidade de 67,8% e especificidade de 98,5% (RATH et al., 2012); enquanto os de segunda, que utilizam tecnologia avançada de fluxo lateral, marcadores fluorescentes e leitores automatizados, apresentam sensibilidade e especificidade de 75,3% e 98,7%, respectivamente (TUTTLE et al., 2015). Apesar do aumento da porcentagem da sensibilidade, é comum a aparição de falsos negativos, sendo essa a principal desvantagem dos ensaios imunocromatográficos (RALSTON et al., 2017).

Em contrapartida, a PCR, descrita em 1980, é uma técnica de alta sensibilidade para a amplificação seletiva de fragmentos de ácidos nucleicos. O método consiste na replicação do fragmento de DNA de interesse por uma DNA polimerase termoestável, gerando uma nova fita que será usada como molde para dar início a um novo ciclo de amplificação. Ao final, após vários ciclos, é possível ter bilhões de cópias do fragmento original (KHAN, 2011). A qPCR é uma variação da técnica que permite amplificação e detecção simultâneas. Devido à sensibilidade, a qPCR é utilizada para diagnóstico de doenças, possibilitando a detecção de patógenos nos estágios iniciais de infecção, quando há uma carga pequena do patógeno, e consequentemente com baixa probabilidade de ser detectada por metodologias menos sensíveis. (WATHUO et al., 2017). No caso do RSV, um vírus de RNA, há uma etapa adicional com a enzima transcriptase reversa, capaz de sintetizar uma fita de DNA complementar (cDNA) ao RNA genômico viral, posteriormente amplificada (ALBERTS et al., 2017b). Embora a PCR

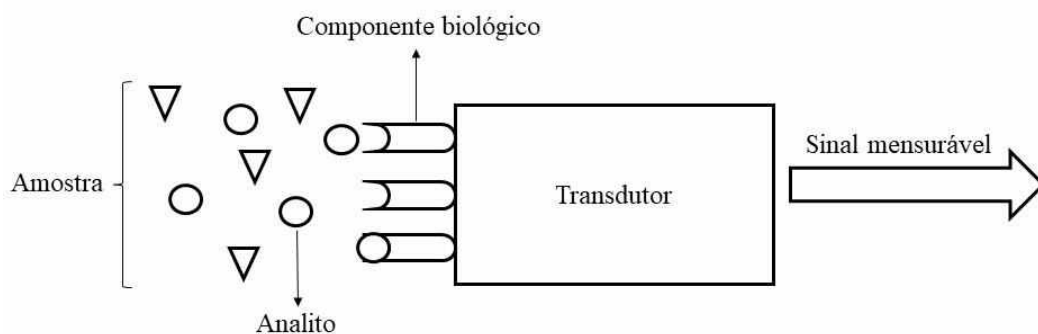


contorne o problema de sensibilidade, é uma técnica que necessita de equipamentos caros e muito específicos, além de levar um tempo maior para obtenção do resultado e precisar de mão de obra qualificada para operação (JAIN; SCHWEITZER; JUSTICE, 2022).

### 2.3 Biossensores

Os primeiros tipos de sensores utilizados foram os físicos, que precisaram ser aprimorados para coletar informações em novos sistemas de gerenciamento, ocasionando no surgimento de sensores químicos e biológicos para suprir essa demanda. Os sensores biológicos ou biossensores são dispositivos capazes de detectar um composto específico ou um grupo seletivo através de uma reação bioquímica entre o analito e um componente biológico acoplado a um transdutor, responsável por converter o sinal da reação em um sinal detectável, proporcional à concentração ou presença do analito (**Figura 4**) (SCHULTZ; TAYLOR, 1996).

**Figura 4** – Constituição de um biossensor.



Fonte: Própria, 2022.

O primeiro biossensor foi desenvolvido em 1962 para a detecção de glicose, utilizando a enzima glicose oxidase acoplada em um transdutor eletroquímico (CLARK; LYONS, 1962), e desde então, são empregados em várias áreas distintas. Na área de alimentos, o sensor biológico é uma alternativa mais simples, barata e rápida para controle de qualidade de um produto do que as técnicas tradicionais, baseadas em análises químicas, espectroscópicas e cromatográficas (MEHROTRA, 2016). Um exemplo é a detecção de tiramina em alimentos fermentados, pois a alta concentração do composto significa a deterioração do alimento devido ao longo período de armazenamento (SOARES et al., 2019).

Biossensores aplicados na área ambiental possibilitam análise *in situ*, economizando tempo e gastos necessários em análises laboratoriais. Esses dispositivos são usados para

controle de qualidade de água, por exemplo, através da detecção de compostos danosos ao meio-ambiente e à saúde humana, como pesticidas (STÉFANNE E SILVA et al., 2020) e corantes da indústria têxtil (VENTURINI ULIANA; YAMANAKA, 2019).

Na área da saúde, os biossensores estão presentes, principalmente, no diagnóstico de doenças infecciosas, como a detecção do vírus Zika por meio da imobilização de RNA no transdutor (DA FONSECA ALVES et al., 2019). Outras aplicações incluem a identificação de pacientes no estágio final de insuficiência cardíaca e a descoberta de novos medicamentos (MEHROTRA, 2016).

### ***2.3.1 Componentes biológicos***

Várias moléculas biológicas podem ser imobilizadas na superfície do transdutor para o desenvolvimento do biossensor. As mais comuns são as enzimas, os ácidos nucleicos e os anticorpos. Outros componentes, como células inteiras e partículas virais também são utilizados, mas em menor proporção.

#### ***2.3.1.1 Enzimas***

As enzimas, catalisadoras capazes de acelerar uma reação em milhões de vezes ou mais, são utilizadas para o desenvolvimento de biossensores, pois apresentam alta especificidade. A reação acontece quando o substrato se liga ao sítio ativo da enzima, formando o complexo enzima-substrato, e posteriormente há formação do(s) produto(s) (ALBERTS et al., 2017c).

O primeiro biossensor enzimático foi também o primeiro biossensor desenvolvido, ou seja, para a detecção de glicose através da imobilização da glicose oxidase (CLARK; LYONS, 1962). Desde então, as enzimas são usadas em sensores biológicos. As vantagens de utilizar uma enzima como componente biológico incluem a especificidade e a afinidade pelo substrato. Por outro lado, a maioria das enzimas são inibidas em certo ponto, quando há uma concentração muito elevada do substrato e/ou produto, além de efeitos no meio, como mudança na temperatura e no pH (SCHULTZ, 1996).

### 2.3.1.2 Ácidos nucleicos

Sensores biológicos de DNA, ou genossensores, utilizam o pareamento de bases entre duas fitas simples da molécula que forma uma fita dupla. Essa interação garante alta especificidade, que aumenta exponencialmente com o aumento de nucleotídeos do fragmento. Sendo assim, há a imobilização de DNA simples fita no transdutor e a detecção é feita a partir da hibridização com DNA complementar presente no analito (SCHULTZ, 1996; WANG, 2000). A detecção do vírus Influenza A é um exemplo de genossensor (DONG *et al.*, 2015).

Apesar do RNA ser encontrado em fita simples (exceção para alguns vírus) devido a sua instabilidade, o ácido nucleico também pode ser usado para desenvolvimento de biossensores, pois possui importante papel na medicina, já que muitos tipos de RNA são biomarcadores, pois sua desregulação é indicativa de doenças, como câncer. Portanto, são moléculas importantes para o desenvolvimento de novas formas de prognóstico e diagnóstico (ISLAM *et al.*, 2017). Labib e colaboradores (2013) desenvolveram um biossensor para detecção de baixos níveis de microRNAs, baseado em hibridização, em ligação proteica e deslocamento de proteína para identificação de células cancerosas, por exemplo.

Os ácidos nucleicos ainda podem formar aptâmeros, fitas com sequências aleatórias com alta afinidade e especificidade de ligação com moléculas alvo, sendo uma alternativa a imobilização de anticorpos devido ao baixo custo, estabilidade, possibilidade de produção *in vitro* e imunogenicidade baixa ou inexistente, podendo ser usados na detecção de bactérias e vírus (LABIB *et al.*, 2012).

### 2.3.1.3 Anticorpos

Os anticorpos são complexos proteicos sintetizados por linfócitos B quando o organismo entra em contato com um corpo estranho, denominado antígeno. Todos os anticorpos apresentam uma estrutura básica com pouca variabilidade, e uma região de ligação do antígeno com enorme variabilidade, demonstrando a capacidade dessas moléculas de se ligarem em uma grande variedade de antígenos estruturalmente distintos. Os anticorpos são compostos por duas cadeias leves e duas cadeias pesadas idênticas, que apresentam regiões aminoterminais variáveis, responsáveis pelo reconhecimento do antígeno, e regiões carboxiterminais constantes, relacionadas às funções protetoras e efectoras desses complexos (ABBAS; LITCHTMAN; PILLAI, 2019a). Na maioria das vezes, os antígenos são moléculas muito maiores do que o sítio de ligação do anticorpo, de forma que possuam regiões específicas de

ligação, denominadas epítomos. A ligação antígeno-anticorpo é irreversível e altamente específica, reconhecendo antígenos que apresentam diferenças mínimas em sua estrutura e composição. Ademais, os anticorpos são amplamente diversos, de modo que um indivíduo consegue produzir grande quantidade desses complexos proteicos, com estruturas distintas (ABBAS; LITCHTMAN; PILLAI, 2019a).

Biossensores baseados em anticorpos, também denominados imunossensores, geralmente utilizam o isotipo IgG imobilizado no transdutor (HOCK, 1997; SCHULTZ, 1996). Devido à grande variabilidade da estrutura desse componente biológico, são aplicados em áreas distintas que vão da ambiental, como para detecção de corantes têxteis em rios (VENTURINI ULIANA; YAMANAKA, 2019), até a da saúde, como para diagnóstico de cânceres (CHAI et al., 2022).

#### 2.3.1.4 Células inteiras

As células são as unidades fundamentais da vida. Todas elas são delimitadas por uma membrana, apresentam meio aquoso em seu interior (citoplasma) e têm capacidade de sintetizarem cópias de si mesmas (ALBERTS et al., 2017a; MADIGAN et al., 2016a). Os primeiros biossensores baseados na imobilização de células inteiras, sendo as procarióticas mais comuns, foram desenvolvidos para coletar informações sobre um estímulo em um organismo vivo, sendo aplicados na farmacologia, na biologia celular, na toxicologia e na área ambiental (BOUSSE, 1996). Atualmente, esses sensores também são usados em análises analíticas, como por exemplo, a imobilização da bactéria *Methylobacterium organophilum* para detecção de metanol (WEN et al., 2014) e a imobilização do fungo *Aspergillus nidulans* modificado geneticamente para detecção de estrogênio (ZUTZ et al., 2017).

#### 2.3.1.5 Outras moléculas

Além das enzimas e anticorpos, outras proteínas que se ligam especificamente a uma molécula alvo são empregadas em sensores biológicos. As lectinas, por exemplo, presentes em sementes de raízes e leguminosas, interagem com carboidratos e são usadas na detecção de glicose, bactérias patogênicas, toxinas e células cancerosas (WANG; ANZAI, 2015). A detecção de *Escherichia coli* por um biossensor baseado na imobilização da lectina concanavalina A, é um exemplo descrito na literatura (GAMELLA et al., 2009).

A imobilização de bacteriófagos, isto é, vírus que infectam exclusivamente células bacterianas, também já foi feita. Richter e colaboradores (2017) demonstraram que bacteriófagos T4 podem ser utilizados para detecção de bactérias, em que *Escherichia coli* foi o microrganismo modelo do estudo.

### 2.3.2 *Transdutores*

Um dos componentes básicos do biossensor é o transdutor, responsável por converter o sinal oriundo da reação bioquímica envolvendo o analito para um sinal detectável. Existem vários tipos de transdutores utilizados para o desenvolvimento de sensores biológicos, de modo que o óptico e o eletroquímico são os mais comuns (SCHULTZ; TAYLOR, 1996).

#### 2.3.2.1 *Transdutores ópticos*

Os transdutores ópticos monitoram as alterações como índice de refração, absorção, reflexão, comprimento de onda, fosforescência e fluorescência na luz no espectro do visível e do infravermelho advindas da reação bioquímica (DAMBORSKÝ; ŠVITEL; KATRLÍK, 2016; LEATHERBARROW; EDWARDS, 1999). Um exemplo é o método de ELISA, usado no diagnóstico de doenças, baseado na interação antígeno-anticorpo e na atividade catalítica de enzimas, como a peroxidase, a fosfatase alcalina e a  $\beta$ -galactosidase. A enzima, ligada ao anticorpo ou ao antígeno, reage com o substrato presente na amostra analisada e acarreta em um produto colorido quantificável, de maneira que a concentração do produto é sempre proporcional à intensidade da cor gerada (MADIGAN et al., 2016b).

#### 2.3.2.2 *Transdutores eletroquímicos*

Biossensores eletroquímicos usam eletrodos que convertem o sinal da reação bioquímica em um sinal elétrico mensurável. Esse dispositivo une a vantagem da imobilização do componente biológico (especificidade) com as vantagens dessa transdução, que englobam sensibilidade e miniaturização. Eles podem ser classificados como amperométricos, potenciométricos, voltamétricos, condutométricos impedimétricos, de acordo com o tipo de sinal elétrico mensurado (WANG, 1996).

Os biossensores amperométricos se baseiam na mensuração de corrente elétrica originada pela oxidação e/ou redução eletroquímica de uma espécie eletroativa. A medida é

possível através do controle de potencial do eletrodo de trabalho (geralmente de platina, de ouro ou de carbono) em relação ao eletrodo referência (geralmente de prata/cloreto de prata). A corrente gerada mede diretamente a taxa da reação de transferência de elétrons, e conseqüentemente é proporcional à concentração do analito de interesse (THÉVENOT et al., 2001; WANG, 1996).

Transdutores potenciométricos convertem o sinal de reconhecimento em um potencial, que, de forma logarítmica, é proporcional a concentração da espécie gerada ou consumida durante a reação. Essa forma de medição proporcionou a expansão do uso de biossensores por não serem baseados no fluxo de elétrons, possibilitando maiores aplicações (WANG, 1996). Nesse caso, uma corrente nula é aplicada para mensurar o potencial entre um indicador e um eletrodo de referência ou dois eletrodos de referência, separados por uma membrana permeável seletiva. Dentre as medidas mais comuns com uso de dispositivos potenciométricos estão pH, além de outros íons ( $F^-$ ,  $I^-$ ,  $Cn^-$ ,  $Na^+$ ,  $K^+$ ,  $Ca^{2+}$  e  $NH_4^+$ ) e gases ( $CO_2$  e  $NH_3$ ) (THÉVENOT et al., 2001).

Biossensores voltamétricos utilizam de técnicas como voltametria cíclica, voltametria de pulso diferencial e voltametria de onda quadrada para a detecção do analito, que ocorre a partir da mudança na corrente e no potencial aplicado, ou seja, há medição da corrente e do potencial. O valor da corrente do pico é usado para identificação do analito, de forma que a densidade da corrente do pico é proporcional à concentração do composto correspondente (SRIVASTAVA *et al.*, 2020).

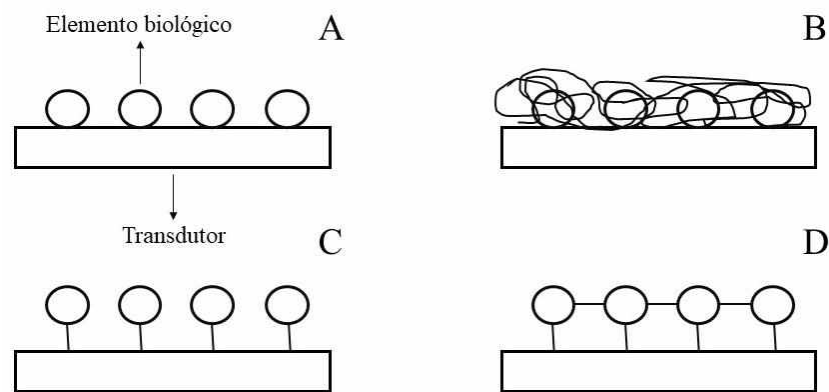
A espectroscopia de impedância eletroquímica (EIS) e a condutometria são duas técnicas relacionadas à medição de propriedades elétricas de soluções, porém possuem diferenças distintas em suas abordagens e aplicações. A condutometria é focada na medição direta da condutividade elétrica de uma solução, relacionada à capacidade de condução de corrente pelos íons presentes nessa solução. É uma técnica simples e direta, amplamente utilizada para determinar concentrações de íons em soluções aquosas. Por outro lado, a EIS é uma técnica mais avançada e abrangente que analisa a resposta elétrica de interfaces eletroquímicas. Ela envolve a aplicação de uma excitação senoidal de frequência variável e a medição da resposta em termos de impedância elétrica em diferentes frequências. A EIS fornece informações detalhadas sobre as propriedades de interfaces eletroquímicas, como resistência, capacitância e reatâncias, permitindo o estudo da cinética de reações, transferência de carga e comportamento de adsorção em sistemas eletroquímicos complexos. Em resumo, enquanto a condutometria é mais simples e voltada para a determinação de concentração de íons, a EIS é

uma técnica avançada que permite uma análise mais detalhada das interfaces eletroquímicas. (BOTT, 2001; GUAN; MIAO; ZHANG, 2004).

### 2.3.3 Técnicas de imobilização

A etapa mais importante no desenvolvimento de um biossensor é a imobilização eficiente do componente biológico na superfície do transdutor. Existem várias técnicas para realizar esse procedimento, cada uma possuindo suas vantagens e desvantagens específicas. A **Figura 5** mostra esquematicamente os principais tipos de imobilização.

**Figura 5** – Técnicas de imobilização do componente biológico no transdutor; (A) adsorção física, (B) oclusão, (C) ligação covalente e (D) *cross-linking*.



Fonte: Própria, 2022.

#### 2.3.3.1 Adsorção física

A adsorção física, mostrada na **Figura 5A**, é a técnica mais antiga e mais simples de imobilização conhecida. Ela é baseada em interações não covalentes entre a superfície do transdutor e o componente biológico, como *Van der Waals*, eletrostática, hidrofóbica e hidrofílica (BHARDWAJ, 2014). Nessa metodologia, geralmente é preparado uma solução do componente biológico que será imobilizado em uma membrana ou filme para possibilitar a adsorção no transdutor. A membrana ou filme pode apresentar grupos iônicos e/ou ter característica hidrofílica ou hidrofóbica, de acordo com as propriedades do componente biológico (TAYLOR, 1996). O primeiro biossensor desenvolvido foi baseado na imobilização da glicose oxidase em uma membrana de polietileno (CLARK; LYONS, 1962). Apesar de sua simplicidade no processo de imobilização, a adsorção física é bastante sensível a mudanças de

pH, de temperatura e de quantidade de íons na amostra, fazendo com que o componente biológico possa ser lixiviado mais facilmente da superfície do transdutor em comparação com outras técnicas (TAYLOR, 1996).

#### 2.3.3.2 *Oclusão*

A oclusão (**Figura 5B**) foi desenvolvida para garantir a imobilização efetiva de moléculas lábeis. Nessa técnica, o componente biológico é preparado juntamente com os constituintes de uma membrana ou filme, ou é difundido após a formação da membrana ou filme na superfície do transdutor, de modo que fique aprisionado, sem ligação química (TAYLOR, 1996). Um exemplo é um dos primeiros biossensores enzimáticos desenvolvidos, o qual houve o oclusão da enzima urease em acrilamida para a detecção de ureia (GUILBAULT; MOLTAVO JR, 1969). Mesmo que esse método consiga aprisionar componentes lábeis através da membrana ou filme, ele consiste em uma interação fraca que pode ser interrompida em condições diversas. Outra desvantagem da oclusão está no fato de que a membrana ou filme pode impedir que o componente biológico interaja com o analito de interesse, pois podem ocupar a região de ligação da molécula (TAYLOR, 1996).

A oclusão pode ser feita por sol-gel, um procedimento de síntese de materiais que transita de um sistema sol, isto é, partículas coloidais com 1-100 nm dispersas de forma estável em um fluido, para um sistema gel, ou seja, uma estrutura rígida formada pelas partículas coloidais ou cadeias poliméricas, de maneira que a parte líquida fique contida nos poros do material (ALFAYA; KUBOTA, 2002). A detecção de galactose pela oclusão de galactose oxidase em um filme de Nafion é um exemplo de biossensor que utiliza o sistema sol-gel (LEE et al., 2021). Outra forma de ocluir o componente biológico é através da formação de um polímero na superfície do transdutor utilizando técnicas eletroquímicas, como por exemplo, a imobilização de glicose oxidase através de um polímero derivado de *m*-fenilenodiamina (ALMEIDA; WINGARD; MALMROS, 1990).

#### 2.3.3.3 *Ligação covalente*

Na ligação covalente (**Figura 5C**), o transdutor tem sua superfície modificada com uma membrana ou filme que possua grupos funcionais disponíveis para que ocorra a ligação química com o componente biológico. A ligação peptídica, responsável pela ligação de aminoácidos para formação de uma proteína (**Figura 6**), é uma forma bastante utilizada de ligação covalente.



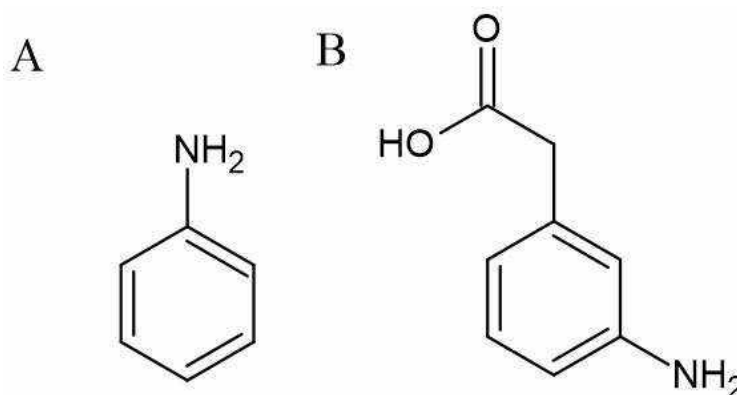


## 2.4 Polímeros condutores

Polímeros são comumente usados para modificar a superfície do transdutor eletroquímico para que a imobilização do componente biológico seja efetiva. Polímeros condutores são compatíveis com moléculas biológicas em solução aquosa de pH neutro e conseguem transferir as cargas elétricas oriundas de reações bioquímicas para um circuito elétrico de forma eficiente (TAHIR; ALOCILJA; GROOMS, 2005). Polímeros condutores combinam propriedades elétricas dos metais e as propriedades mecânicas dos polímeros. Esses materiais apresentam uma cadeia principal com ligações simples ( $\sigma$ ) e ligações duplas ( $\pi$ ) alternadas. As ligações simples são fortes e possuem um maior número de elétrons localizados, enquanto as ligações duplas são fracas e possuem menos elétrons localizados, e consequentemente, permitem uma maior mobilidade. Entretanto, para que um polímero seja efetivamente condutor, é necessário que os elétrons sejam removidos por dopagem, formando vacâncias, que são preenchidas por outros elétrons da estrutura, gerando novas vacâncias (KUMAR; SHARMA, 1998; MEDEIROS et al., 2012).

Polímeros condutores provenientes de monômeros que apresentam anel aromático, como a anilina, melhoram a performance do biossensor e impedem a adsorção de interferentes na superfície do transdutor (**Figura 7A**), que gerou o primeiro polímero condutor a ser comercializado e já possui sua eletropolimerização muito bem estudada devido ao grande número de aplicações, incluindo o desenvolvimento de biossensores eletroquímicos (DHAND et al., 2015).

**Figura 7** – Estrutura química da (A) anilina e do (B) 3APA.



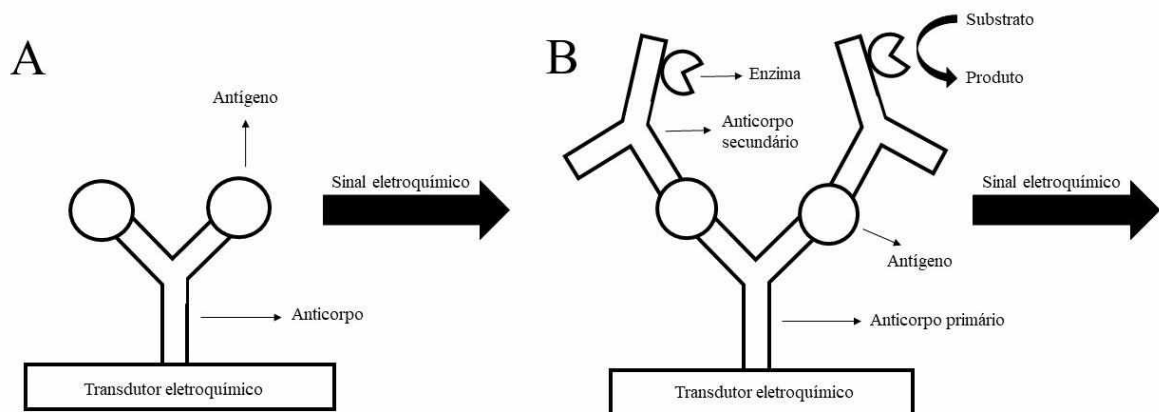
Fonte: Própria, 2022.

Monômeros derivados da anilina são possíveis candidatos a formarem polímeros condutores, como o ácido 3-aminofenilacético (3APA) (**Figura 7B**), com fórmula molecular  $C_8H_9NO_2$  e peso molecular  $151,16 \text{ g.mol}^{-1}$ , que apresenta um grupo carboxila, um grupo amino e um anel aromático (NATIONAL CENTER FOR BIOTECHNOLOGY INFORMATION, 2022), características encontradas em monômeros com aplicabilidade no desenvolvimento de biossensores. O grupo carboxila, não ligado diretamente ao anel aromático, normalmente não é suscetível à eletropolimerização, ou seja, fica disponível para a ligação covalente com o elemento biológico. (ALVES-BALVEDI et al., 2016). O monômero apresenta um grupo amino ligado diretamente ao anel aromático, que está envolvido no mecanismo de ação de sua polimerização eletroquímica.

## 2.5 Imunossensores eletroquímicos

Os imunossensores eletroquímicos exploram a ligação antígeno-anticorpo para detecção, de modo que um deles seja imobilizado na superfície do transdutor. A detecção do analito pode ser feita de forma direta (**Figura 8A**), através da medida eletroquímica gerada pela ligação antígeno-anticorpo, ou, após a interação entre o componente biológico imobilizado e o analito, havendo um segundo anticorpo, marcado com enzima, incorporado ao sistema, juntamente com um substrato enzimático adequado (**Figura 8B**). Dessa forma, a reação enzimática gera um produto que fornece uma medida eletroquímica (RICCI; ADORNETTO; PALLESCHI, 2012).

**Figura 8** – Componentes básicos de um imunossensor eletroquímico com detecção (A) direta e (B) com anticorpo secundário marcado.



Fonte: Própria, 2022.

A performance de polímeros condutores em imunossensores pode ser vista no trabalho de Lorenzen e colaboradores (2022), através da imobilização da nucleoproteína do vírus SARS-CoV-2 para diagnóstico de COVID-19, pela detecção do anticorpo na amostra de pacientes. Desse modo, o eletrodo de trabalho composto por aço foi modificado com polímero derivado de 3,4-etilenodioxifenol e nanopartículas de ouro. Outro trabalho, desenvolvido por Palomar (2018) e colaboradores também utilizou um polímero condutor para melhorar resposta em um biossensor impedimétrico para detecção de anticorpos de dengue. Para isso, foi usado eletrodo de carbono vítreo modificado com nanotubos de carbono e polímero condutor derivado de pirrol, para a imobilização da proteína NS1 do vírus da dengue.

## CAPÍTULO 2: ARTIGO CIENTÍFICO

### Development of a label-free impedimetric immunosensor for the detection of respiratory syncytial virus

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**Abstract:** Respiratory syncytial virus (RSV) is an important pathogen responsible for cases of bronchiolitis, bronchitis, and pneumonia, worryingly effective and lethal to children under the age of five. Current detection methods for RSV are expensive, demand time, and high-skilled personnel. Therefore, in this work, we developed a cost-effective impedimetric immunosensor for the detection of RSV based on a pencil graphite electrode (PGE) modified with a polymeric film derived from 3-amino phenylacetic acid (3APA). The polymer was formed and characterized through cyclic voltammetry. The current response decrease in potassium ferricyanide solution and the current response increase in methylene blue solution suggest the presence of unchanged carboxylic groups. The number of electrons involved in the first step of the reaction and the electron/proton ratio were like the one proposed for aniline, and a mechanism of electropolymerization was proposed. The immunosensor was assembled by the activation of the carboxyl groups, plunging the modified electrode in an EDC/NHS solution, followed by a nucleophilic substitution reaction with the amino groups present in the anti-RSV antibody (Ab). The surface was blocked with glycine before the incubation with the virus. Detection was performed using electrochemical impedance spectroscopy (EIS), and the data were evaluated using equivalent circuits. The values of resistance of charge transfer ( $R_{ct}$ ) increased as the PGE was modified in the following order: PGE < PGE/polymer < PGE/polymer/Ab < PGE/polymer/Ab/virus. For optimization, 100 ng of Ab was used, and the immobilization time for Ab and Gly were: 3 h and 50 min, respectively. The response time of the system, with Ag immobilization, was 30 min. The limit of detection was  $27.65 \text{ PFU.mL}^{-1}$ , and the limit of quantification was  $92.15 \text{ PFU.mL}^{-1}$ , with a linear range from 50 – 1000  $\text{PFU.mL}^{-1}$ , demonstrating that the proposed immunosensor is a viable alternative for the detection of RSV.

**Keywords:** Respiratory syncytial virus, biosensor, electropolymerization

## 1. Introduction

Acute respiratory infections are characterized by the obstruction of air passage in the respiratory system, and it is one of the major causes of illness and death, especially in children. The respiratory syncytial virus (RSV) is one of the most important pathogens responsible for this scenario, especially in children up to five years old. It is characterized by failure to develop long-term immunity, leading to lifelong reinfections. In 90% of cases, the first contagion occurs until the first two years of life [1], and in severe cases, it can lead to pneumonia and bronchiolitis [2].

It is a contagious virus, as it enters the host organism by the conjunctiva-upper respiratory tract mucous through direct contact or contaminated droplets. After the incubation time (2-8 days), it spreads until it reaches the epithelium, where its growth is favored [1,3]. Roughly 30 million new cases of RSV infection occur annually, with more than 100 thousand deaths of children. Developing countries present a higher mortality rate, probably because of the late diagnosis, corresponding to 90 % of all deaths caused by the virus [4].

Generally, diagnosis is performed through physical exams based on the patient's symptoms. However, because of the high number of different microorganisms that can generate the same symptoms, the precise identification of RSV is only possible through laboratory analysis. The indirect immunofluorescence assay is the most common method with high sensitivity [5], the immunochromatographic assays, also known as rapid tests, provide results within a 10 to 30-minute range and are cost-effective [6], and the RT-qPCR is the gold standard method, in substitution of the virus isolation in cell culture [7]. However, there is a demand for a method that can rapidly determine the virus specifically with high sensitivity and cost-effectiveness in a friendly-used device, all advantages usually found in biosensors.

The structure and physiopathology of RSV are well-settled in the literature [8], as well its immunological response [9], with a library of specific antibodies available as the pivotal material for the effective response of the immunoassays and strong candidates for an immunosensor development. Electrochemical impedance spectroscopy (EIS) is a feasible choice and a powerful tool [10] for label-free RSV determination based on the strong antibody-antigen interaction through interfacial properties changes, and it has been useful for biosensor development [11–13]. The efficient biomolecule immobilization over the transducer surface is a key-step in the device assembly, and functionalized polymers

electrogenerated over electrodes are fulfilling this requirement by covalently attaching the biomolecules through the available functional groups [14–16]. In this sense, the monomer 3-amino phenylacetic acid (3APA) is an excellent candidate as a polymeric matrix. The carboxylic acid, away from the ring by a methyl group, tends to remain unchanged and free to bind to the biomolecules through a covalent bond, while it is expected that the polymerization begins from the nitrogen atom like the well-known mechanism for polyaniline [17].

Therefore, this work focused on the electropolymerization of 3APA over pencil graphite electrodes (PGE) and the characterization of the formed product, followed by the development of a label-free impedimetric immunosensor for RSV determination. To the best of our knowledge, this is the first attempt to develop an impedimetric immunosensor for RSV using this polymeric matrix over PGE.

## 2. Materials and methods

### 2.1. Chemicals

All solutions were prepared using deionized water, and all the chemicals were used as received. Sulfuric acid ( $\text{H}_2\text{SO}_4$ ) was obtained from LS Chemicals®, potassium chloride (KCl) and potassium dihydrogen phosphate ( $\text{KH}_2\text{PO}_4$ ) from Dinâmica®, potassium ferricyanide ( $\text{K}_3\text{Fe}(\text{CN})_6$ ), boric acid ( $\text{H}_3\text{BO}_3$ ), and sodium hydrogen phosphate ( $\text{Na}_2\text{HPO}_4$ ) from Vetec®, methylene blue ( $\text{C}_{16}\text{H}_{18}\text{ClN}_3\text{S}$ ) and dimethylsulfoxide (DMSO) from Isofar®, 3APA, 2-methanesulfonic acid (MES) ( $\text{C}_6\text{H}_{13}\text{NO}_4\text{S}$ ), EDC ( $\text{C}_8\text{H}_{17}\text{N}_3$ ), and NHS ( $\text{C}_4\text{H}_5\text{NO}_3$ ) from Sigma-Aldrich®, acetic acid ( $\text{CH}_3\text{COOH}$ ) and phosphoric acid ( $\text{H}_3\text{PO}_4$ ) from Labsynth®, sodium hydroxide (NaOH) from Synth®, Anti-RSV Antibody, fusion protein, all type A, B strains, clone 131-2A from Merck Millipore® (MAB8599), sodium chloride (NaCl), Glycine ( $\text{C}_2\text{H}_5\text{NO}_2$ ), and sodium bicarbonate ( $\text{NaHCO}_3$ ) from Êxodo Científica®, fetal bovine serum (FBS), and *Dulbecco's* Modified Eagles (DMEM) medium from Gibco®, penicillin  $10000 \mu\text{g}\cdot\text{mL}^{-1}$  and streptomycin  $10000 \text{U}\cdot\text{mL}^{-1}$  from Hyclone®.



## 2.2. Equipment

Cyclic voltammetry (CV) and electrochemical impedance spectroscopy (EIS) experiments were performed in a potentiostat model PGSTAT204 with an FRA module from Metrohm (Netherlands). PGE mines from Pentel, hardness HB, named Super Hi-Polymer C505, with a diameter of 0.9 mm, were used as working electrodes. To ensure reproducibility, the area of the electrode was kept constant at 0.289 cm<sup>2</sup> for all experiments. Platinum wire and Ag/AgCl (KCl 3.0 M) electrodes were used as auxiliary and reference electrodes. The PGEs were mechanically polished using sandpaper, followed by washing in deionized water, ultrasonication, and drying using ultra-pure N<sub>2</sub> gas.

## 2.3. Electropolymerization and characterization

The electropolymerization of 3APA (1.0 mM) was performed using sulfuric acid (0.10 M) as a supporting electrolyte through CV (0.05 V s<sup>-1</sup>). The voltammograms were recorded from 0.0 to 1.2 V. The system was analyzed through CV in KCl (0.5 M), K<sub>3</sub>Fe(CN)<sub>6</sub> (5,00 mM) in KCl (0,50 M), C<sub>16</sub>H<sub>18</sub>ClN<sub>3</sub>S (5,00 mM) in KCl (0,50 M), and in H<sub>2</sub>SO<sub>4</sub> (0,10 M) solutions to evaluate the modification of the electrode and to characterize the material. EIS measurements were performed in a K<sub>3</sub>Fe(CN)<sub>6</sub> (5.00 mM) in KCl (0.50 M) solution, frequency range from 1x10<sup>5</sup> to 1x10<sup>-2</sup> Hz, 1x10<sup>-2</sup> V<sub>RMS</sub>; frequency number = 60; open circuit potential.

## 2.4. Virus and cell culture

Respiratory syncytial virus, subgroup A, strain A2, were obtained from Laboratório de Cultura de Células Animais, Universidade Federal de Uberlândia, campus Patos de Minas. The virus was propagated in HEp-2 cells (ATCC CCL-23) in Dulbecco's modified Eagle medium (DMEM) free of antibiotics and fetal calf serum (FCS), and the supernatant was used in the titration and Immunoelectrochemical assay. Culture supernatant from un-infected HEp-2 cells was processed similarly to use for mock virus preparation. 0,5x10<sup>7</sup> PFU mL<sup>-1</sup> of RSV titration was determined in plaque forming units (PFU) assay. Subsequently, the virus was inactivated by heating at 100 °C for 15 min and was stored at -80 °C for further use.

### 2.5. Immunosensor assembly and RSV determination

Anti-RSV and RSV stock solutions ( $50 \text{ ng mL}^{-1}$  and  $5.00 \times 10^6 \text{ PFU mL}^{-1}$ , respectively) were diluted in PBS 1x, pH 7.4 before each experiment. The carboxyl groups of the polymer were first activated, plunging the modified electrode in an EDC (0.1 M) and NHS (0.02 M) in MES (0.1 M, pH 4.5) aqueous buffer solution for 30 minutes. After, the electrode was plunged in a  $15.0 \text{ }\mu\text{L}$  diluted antibody solution ( $6.67 \text{ ng }\mu\text{L}^{-1}$ ) for 3 hours. The sensor surface was blocked with glycine (50.0 mM) in PBS 1x, pH 7.4, for 50 minutes, as previously reported [18]. Lastly, the immunosensor was plunged into a  $15.0 \text{ }\mu\text{L}$  diluted RSV solution for 30 minutes. Washing was performed after each step. The RSV was determined through EIS in the same configuration presented in section 2.3.

### 2.6. Optimization and validation

The impedimetric immunosensor was optimized ( $n=3$ ) regarding the amount of antibody, time of antibody, glycine, and RSV incubation. The limit of detection (LOD) was calculated using equation  $3s_b/a$ , and the limit of quantification (LOQ) was calculated using the equation  $10s_b/a$ , where  $s_b$  is the standard deviation of a blank solution, and  $a$  is the slope of the linear equation obtained from the calibration curve at different RSV concentrations. As the virus was obtained from cell culture, cell culture in the absence of the virus was tested to discard interference of the matrix on the positive response.

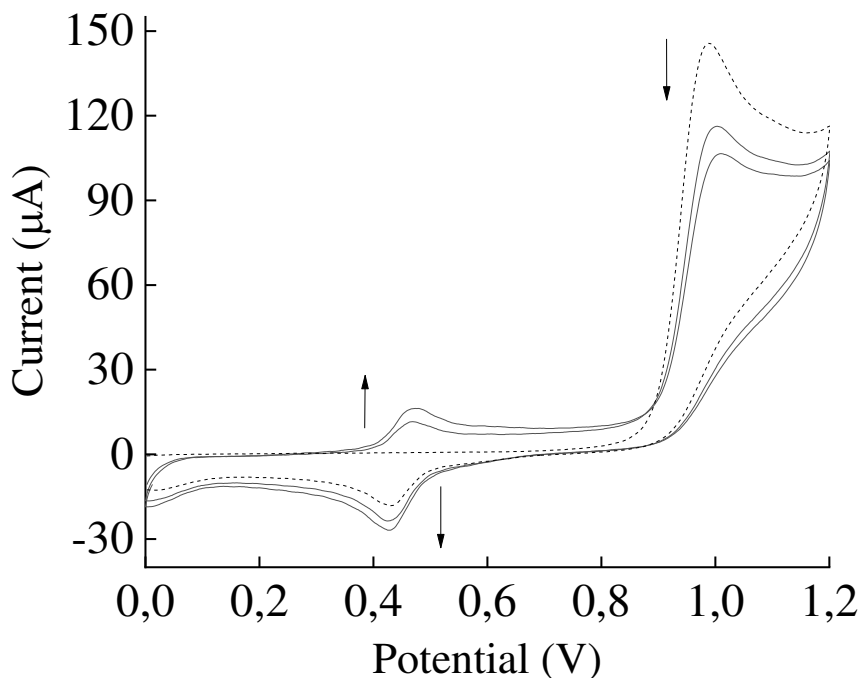
## 3. Results and discussion

### 3.1 Electropolymerization and characterization

The first three scans obtained in a 3APA/sulfuric acid solution over PGE performed through CV resulted in a faradaic anodic response at  $+0.99 \text{ V}$ , followed by a redox couple signal at  $+0.43/+0.48 \text{ V}$  (**Fig.1**). The first signal results from an irreversible monomer oxidation, expected to occur on the nitrogen atom to generate the first radical cation. The current response decreases with subsequent scans with a slight shift to a more anodic potential due to the surface modification with a material hampering the diffusion

of new monomer species toward the electrode. The redox couple presents a current increase with subsequent scans, suggesting a material growth over the electrode surface. These responses at a more cathodic potential show the easiness in the electron transference to and from the electrode, as expected from novel aromatic structures with more mesomeric forms. This analysis is under the classic responses obtained from aniline [19] and derivatives [20,21] electropolymerization in acid conditions.

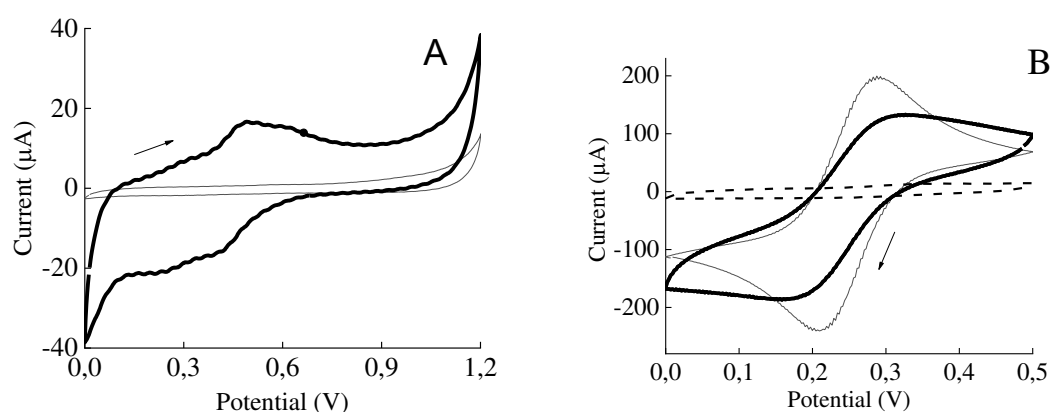
**Fig.1.** CVs recorded in 3APA (2.5 mM) using H<sub>2</sub>SO<sub>4</sub> (0.5 M) as supporting electrolyte over PGE, 50 mV s<sup>-1</sup>, 3 cycles. The arrows indicate the increase/decrease of the current response. The first scan is represented by a dashed line.

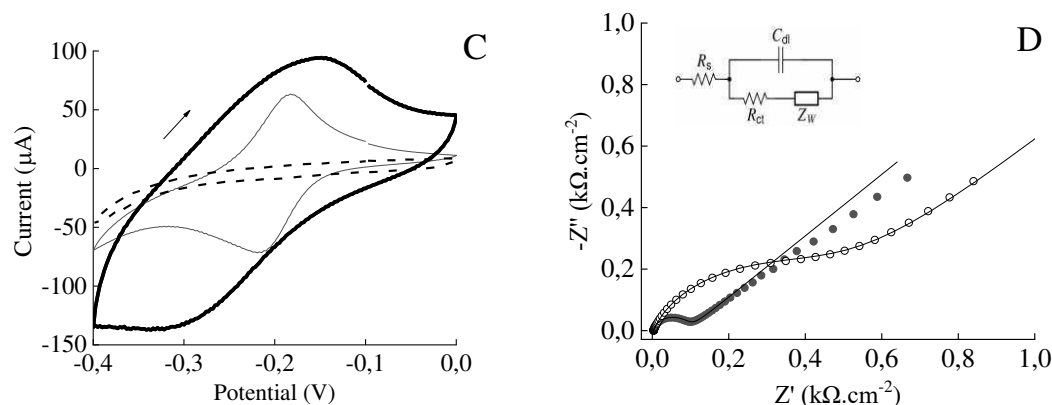


The electrode modification was first evaluated in the blank supporting electrolyte solution (**Fig.2A**), where it is possible to confirm the absence of the monomer oxidation signal at more anodic potential and the persistence of the redox couple current signal. The presence of the carboxyl groups in the material can be assessed by the voltammograms performed in a ferricyanide solution and a methylene blue (MB) solution. The KCl aqueous solutions present a pH value of around 6-7, in which the carboxylate anions are predominant. Therefore, it is expected an electrostatic repulsion between carboxylate and ferricyanide, as well as an electrostatic attraction between the charged polymer and MB. The current decrease observed in **Fig.2B** and the current increase observed in **Fig.2C** strengthen the repulsion/attractive hypothesis, suggesting the presence of the functional

group alongside the polymeric chain, which is essential for biosensor development. The Nyquist diagram presented in **Fig.2D**, and the experimental EIS data fitted through the Randles circuit (**Table 1**), corroborate the data obtained in the ferricyanide solution with a 5.32 times increase in  $R_{ct}$  values from PGE ( $89.68 \Omega \text{ cm}^{-2}$ ) to the modified PGE ( $477.20 \Omega \text{ cm}^{-2}$ ). The  $R_s$  are practically the same for both systems, showing good reproducibility. There is also a 19.86 times increase in  $Q_{dl}$ , suggesting the presence of a more compact material over PGE, filling the graphite gaps and not only the outer shell. Moreover, this difference might be related to the amount of charge present in the electrode surface, which could be resulted from the charged carboxylate anions. The Warburg resistance is similar between the systems, which indicates that the mass transport is unaltered with the electrode modification. The electrode modification was optimized with the number of cycles. It was observed that 30 cycles are an optimum number of cycles based on the small differences in the electrochemical profile performed in potassium ferricyanide and sulfuric acid solutions from this point on. Also, the system was kept stored at  $-4^\circ\text{C}$ . After ten months, the modified electrode retained 91 % (anodic current) and 83 % (cathodic current) of its initial ferricyanide response, which proves that the material is stable with storage time and can be used for further biosensor development.

**Fig.2.** CVs recorded in (A)  $\text{H}_2\text{SO}_4$  (0.5 M) at  $50 \text{ mV s}^{-1}$ , (B)  $\text{K}_3\text{Fe}(\text{CN})_6$  (5.00 mM) in  $\text{KCl}$  (0.50 M) at  $100 \text{ mV s}^{-1}$ , (C) methylene blue (5.00 mM) in  $\text{KCl}$  (0.5 M) at  $100 \text{ mV s}^{-1}$ , and (D) Nyquist diagram recorded in  $\text{K}_3\text{Fe}(\text{CN})_6$  (5.0 mM) containing  $\text{KCl}$  (0.1 M). OCP at 0.25 V;  $\Delta E = 10 \text{ mV}$ ; Frequency range: 100,000 Hz to 10 mHz over PGE (line/open circle for D), and poly-(3APA)-modified PGE (bold lines, gray circles in D). Dashed lines in B and C represent measurements recorded in  $\text{KCl}$  (0.5 M) for PGE. Solid lines in D represent the fit of the equivalent circuit to the experimental findings. Inst in figure D shows the Randles circuit used to fit the data. The arrows indicate the initial scan direction.





**Table 1.** Parameters obtained from EIS data simulation for PGE and modified PGE using Randles circuit.

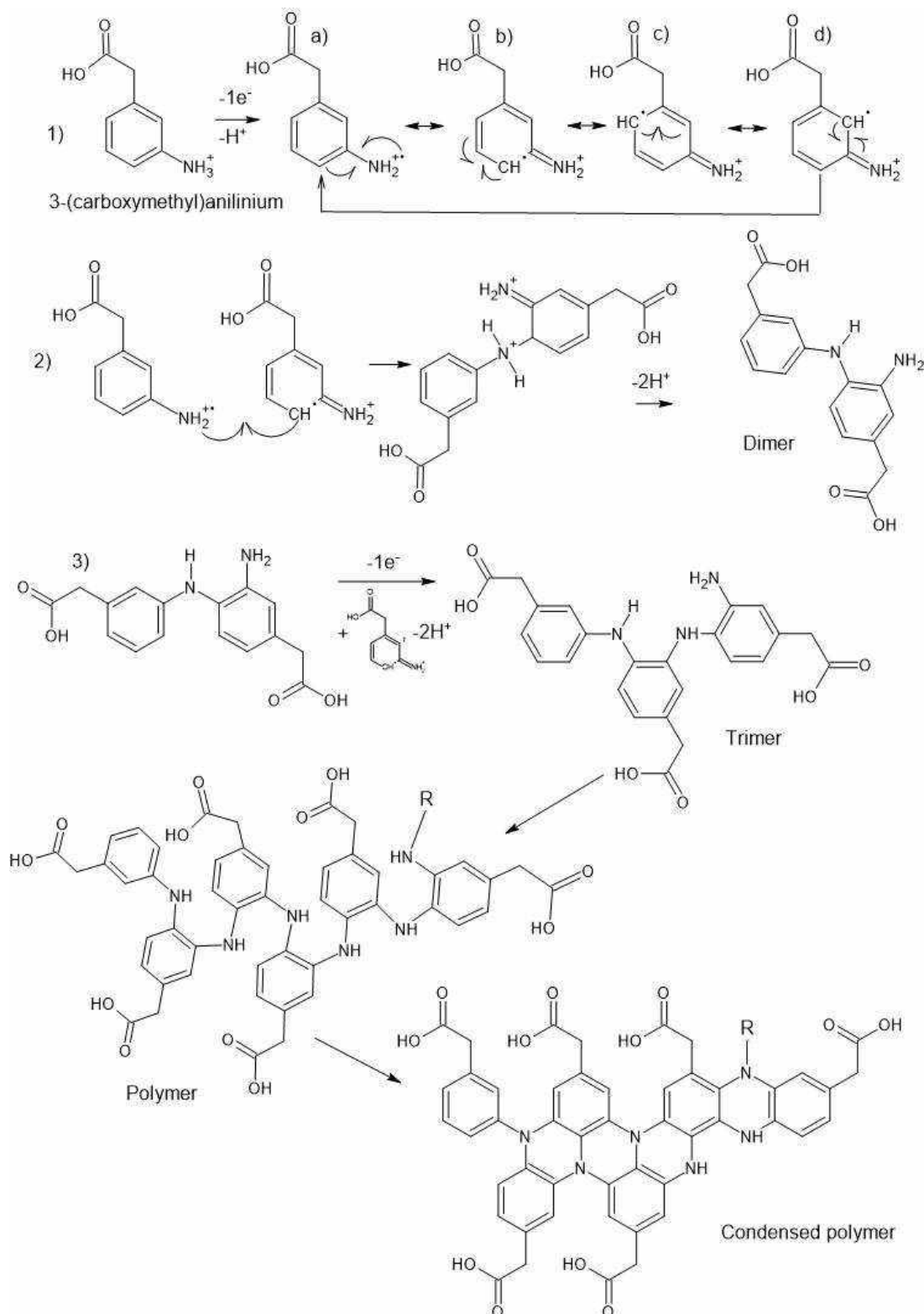
Parameter	PGE	PGE/poly-(3APA)
$R_s$ ( $\Omega$ )	11.36	12.01
$R_{ct}$ ( $\Omega.cm^{-2}$ )	89.68	477.20
$Q_{dl}$ ( $m\Omega.s^N$ )	$6.67 \times 10^{-6}$	$1.32 \times 10^{-4}$
N	0.93287	0.79756
$Z_w$ ( $m\Omega.s^{1/2}$ )	$1.42 \times 10^{-3}$	$1.16 \times 10^{-3}$
$\chi^2$	0.0115	0.0423

### 3.2 Mechanism proposal

Electrochemical equations can provide useful information to understand the mechanism of 3APA electropolymerization, as was previously reported [14]. Cyclic voltammograms were recorded in the acidic monomeric solution changing the scan rate. Linear regression from plots of the log of peak current vs peak potential (**Fig.S1A**) and peak potential vs log of scan rate (**Fig.S1B**) were applied. The Tafel and Laviron equations were used based on the slope obtained from each study, thus providing the electron transfer coefficient ( $\alpha$ ) and the number of electrons ( $n$ ), respectively. For an  $\alpha$  of 0.445 for 3APA, the number of electrons calculated was 1.55. Because of the similarity of the monomer with aniline, it is more likely that the first step of the reaction involves one electron than two. Monomer solutions were also prepared in BR buffer at different pH values, and a voltammogram of the first scan was recorded in each solution. The peak potential vs pH profile was obtained (**Fig.S2**) with a linear behavior between pH 2 and 5 followed by a plateau, indicating no structural changes after this point, which matches the second monomer pKa. According to the Nernst equation [22], with a linear slope of 0.05598, it is believed that the monomer's first oxidation occurs with the involvement of

the same number of protons and electron, more precisely, one proton and one electron, based on the previous study. Thus, with the information gathered through the electrochemical experiments, an electropolymerization mechanism was proposed (**Fig 3**).

**Fig.3.** Electropolymerization mechanism proposal for poly-(3APA).



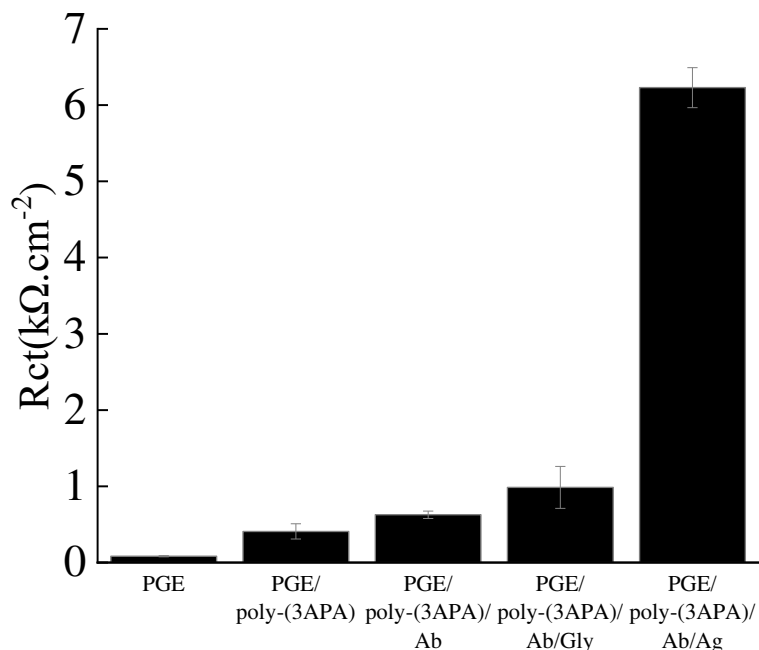
In strong acidic media, 3APA is protonated in the form of 3(carboxymethyl)anilinium cation. Step 1 is the electrochemical withdrawal of one electron and one proton from the ammonium group to generate the radical cation stabilized in 4 mesomeric forms (a to d). Among them, structure (b) presents less steric hindrance compared to structure (c) because of the proximity of the radical with the carboxyl group and to structure (d) with the radical between two bulky groups. Step 2 is the homolytic reaction between structures (a) and (b), followed by the withdrawal of two protons by the solvent, regenerating an aromatic in the form of a neutral dimer. This structure is believed to be responsible for the redox couple response during electropolymerization at less anodic potentials because of the easiness of electron removal from a structure with more mesomeric forms. After subsequent reactions (step 3), a polymer is formed. Here, we present two possible polymer structures. Based on the proximity of the nitrogen atom to the aromatic ring, it can be expected a formation of condensate rings, improving the material stability. Regardless of the final structure, both present the same desired array, with the unchanged carboxyl group, which will be used for further reaction with the antibody.

### 3.3 Biosensor development

EIS was performed after each step of the biosensor construction, and the data was fitted using the Randles circuit. The  $R_{ct}$  was used as the parameter of analysis for all the experiments (**Fig 4**). It is expected an increase in the values of resistance of charge transfer after each electrode modification. The resistance increase from the polymer ( $477.20 \Omega \text{ cm}^{-2}$ ) to the antibody ( $642.16 \Omega \text{ cm}^{-2}$ ) suggests that the biomolecule was efficiently immobilized through the covalent bond. Even though antibodies are large molecules, the resistance increase is not substantial because we only used 100 ng of the biomaterial. The resistance increases with further surface blocking with glycine ( $957.20 \Omega \text{ cm}^{-2}$ ), avoiding non-specific adsorption over the electrode surface. Then, a 1.8 times resistance increase ( $1727.27 \Omega \text{ cm}^{-2}$ ) was observed with the virus, proving the biosensor efficiency in the presence of the antigen.

**Fig.4.** Column chart of  $R_{ct}$  obtained through fitting using the Randles circuit from EIS data recorded in  $\text{K}_3\text{Fe}(\text{CN})_6$  (5.0 mM) containing KCl (0.1 M). OCP at 0.25 V;  $\Delta E = 10$  mV; Frequency range: 100,000 Hz to 10 mHz. Poly-(3APA)-modified PGE with 30 cycles in a 3APA (2.5 mM) in  $\text{H}_2\text{SO}_4$  (0.5 M) solution,  $50 \text{ mV s}^{-1}$ , 100 ng Ab (6.67 ng

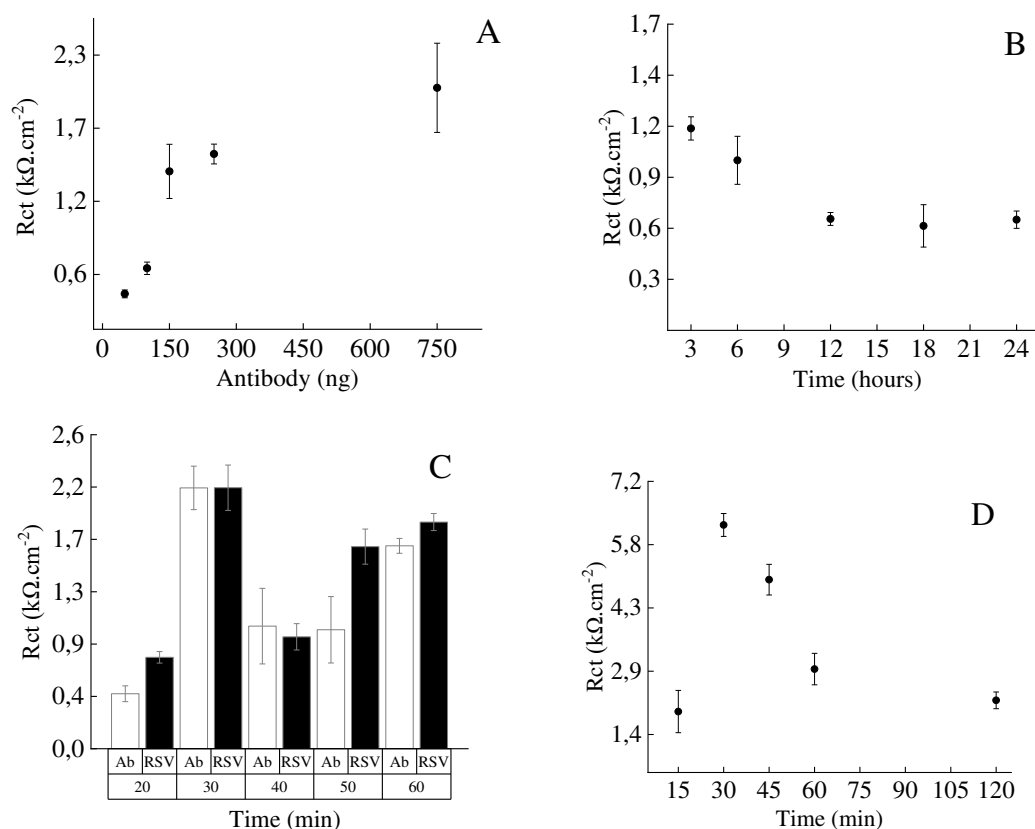
$\mu\text{L}^{-1}$ ) in PBS 1x pH 7.4 with an incubation time of 3 h, gly = glycine (50.0 mM) in PBS 1x pH 7.4 with an incubation time of 50 min, and VSR ( $5.0 \times 10^4$  PFU  $\text{mL}^{-1}$ ) in PBS 1x pH 7.4 with incubation time of 30 minutes.



The biosensor was optimized regarding the amount of antibody, the incubation time with the antibody, glycine, and the virus (**Fig 5**). From **Fig.5A**, it is possible to see that the  $R_{ct}$  increase is proportional to the antibody concentration. The two extreme concentration values are not interesting for further analysis. The  $R_{ct}$  value for antibody concentration of 50 ng is below the same value obtained for the polymer. At the other end, the  $R_{ct}$  for the antibody at 750 ng is somewhat similar to the response obtained at 250 ng. From this sole point of view, one would consider using the 250 ng for further experiments. However, the proper biosensor measurement is based on the differentiation between the Ab-Ag response from the Ab response, and the high Ab concentration might not be suitable from this aspect. Therefore, experiments with Ag must be performed to assess the highest difference between the  $R_{ct}$  values. **Fig.S3** presents this data. The value of 250 ng does not provide a good differentiation between the responses. The higher differences come from 100 and 150 ng. We chose 100 ng based on the high reproducibility obtained by comparing the error bars from the two experiments. This choice also results in cost-effectiveness with less expensive biomaterial used, especially considering further mass-scale production for a point-of-care device.



**Fig.5.** Optimization studies, all starting from PGE/poly-(3APA). All steps were prepared in PBS 1x pH 7.4 solution.  $R_{ct}$  obtained through fitting using the Randles circuit from EIS data recorded in  $K_3Fe(CN)_6$  (5.0 mM) containing KCl (0.1 M). OCP at 0.25 V;  $\Delta E = 10$  mV; Frequency range: 100,000 Hz to 10 mHz. (A) Ab amount with an incubation time of 24 h; (B) Ab incubation time, 100 ng Ab ( $6.67 \text{ ng } \mu\text{L}^{-1}$ ); (C) Glycine (50.0 mM) incubation time over PGE/poly-3(APA)/Ab and after immobilization of RSV ( $5.0 \times 10^4 \text{ PFU mL}^{-1}$ ) with incubation time of 3 hours; and (D) RSV( $5 \times 10^4 \text{ PFU mL}^{-1}$ ) incubation time.

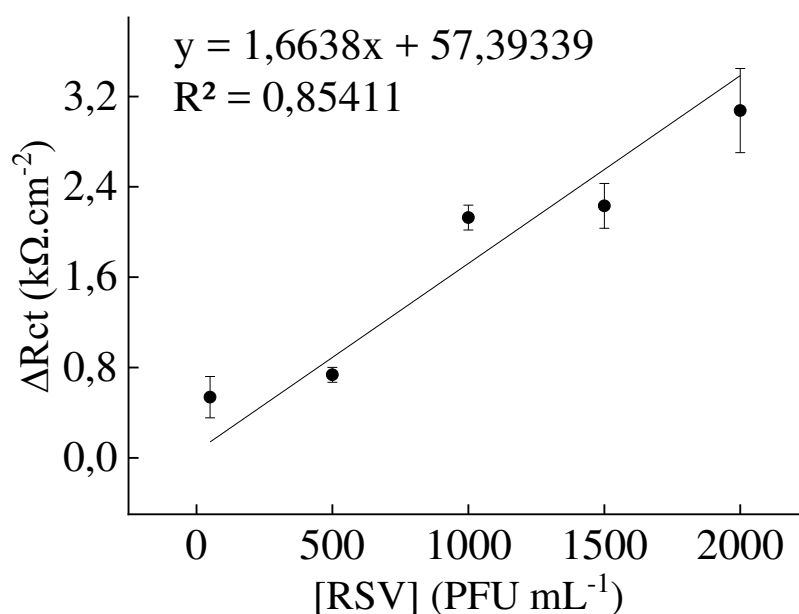


The  $R_{ct}$  response decreases with Ab incubation time (**Fig.5B**), reaching a plateau after 12 hours. Experiments were performed under 3 hours of incubation. However, it was not possible to interpret the results or perform a fitting using the equivalent circuit since the data were all scattered in a non-reproducible way. This condition might be connected with the EDC/NHS presence, which requires more time for the chemical reaction with the biomolecule to occur. It is worth mentioning that the electrodes are plunged into microtubes with the carboxylic acid activation solution and not otherwise (drop-casting the solution over the electrode). The interaction time by plunging is naturally higher than the drop-casting method because of the slow diffusion of the molecules toward the electrode surface. Acceptable Nyquist diagrams were only possible after 3 hours of incubation, with a reproducible response. The  $R_{ct}$  values decrease after this time is related

to the saturation of the electrode surface [23]. Therefore, 3 hours were chosen as the Ab incubation time.

The time for glycine incubation was also evaluated as the surface-blocking agent [18,24,25]. The best  $\Delta R_{ct}$  for glycine incubation time (**Fig.5C**) was 50 minutes. Lastly, the incubation time for the virus was evaluated (**Fig.5D**). It was observed a maximum response at 30 minutes, similar to other reported works regarding respiratory viruses [26,27]. The optimized biosensor was applied in a calibration curve using different RSV concentrations. The linearity was observed from 50.0 and 2000.00 PFU mL<sup>-1</sup> (**Fig.6**) with a sensitivity of 1.6638 PFU mL<sup>-1</sup>. The calculated LOD and LOQ were 22.54 and 75.12 PFU mL<sup>-1</sup>, respectively. The sensor also presents reproducibility with an RSD of 9.38 % (n = 10). **Table 2** contains the data from other published papers regarding the LOD obtained for RSV determination. The proposed biosensor presented lower LOD compared to other electrochemical systems [28,29], similar values compared to the optic sensors [30–34], and specifically to PCR assays [33,35,36]. It is a remarkable feature of a label-free, cost-effective system without any support from the use of nanoparticles.

**Fig.6.** Calibration curve (50.0 – 2000 PFU mL<sup>-1</sup>) obtained through EIS for different RSV concentrations recorded in K<sub>3</sub>Fe(CN)<sub>6</sub> (5.0 mM) containing KCl (0.1 M). OCP at 0.25 V;  $\Delta E = 10$  mV; Frequency range: 100,000 Hz to 10 mHz. Poly-(3APA)-modified PGE with 30 cycles in a 3APA (2.5 mM) in H<sub>2</sub>SO<sub>4</sub> (0.5 M) solution, 50 mV s<sup>-1</sup>, 100 ng Ab (6.67 ng  $\mu$ L<sup>-1</sup>) with an incubation time of 3 h, gly = glycine (50.0 mM) with an incubation time of 50 min, incubation time of RSV of 30 minutes.



**Table 2** – LOD obtained from other different sensors for RSV detection.

<b>Detection method*</b>	<b>LOD (PFU mL<sup>-1</sup>)</b>	<b>Ref</b>
Electrochemical – EIS	1100.00	[28]
Electrochemical – Potentiometric	1000.00	[29]
Optic – Fluorescence	595.00	[30]
Optic – SERS	100.00	[31]
Optic – ELISA	50.00	[32]
Optic – ELISA	16000.00	[33]
Optic – LSPR	40.00	[34]
RTrtPCR	14.13	[35]
TaqMan RT-PCR	1.80-2.30	[36]
RT-qPCR	17.90	[33]
NPA-IPCR	4.10	[33]
Electrochemical – EIS	22.54	This work

\*SERS: Surface-enhanced Raman spectroscopy, LSPR: Localized Surface Plasmon Resonance, PCR: Polymerase Chain Reaction, RTrtPCR: real-time Reverse Transcriptase PCR, RT-qPCR: Quantitative Reverse Transcriptase PCR; NPA-PCR: Nanoparticle Amplified Immuno-PCR

## Conclusions

A stable polymer derived from 3APA was electrochemically formed and characterized over PGE, resulting in a material with a functionalized carboxyl group. The covalent bond through a nucleophilic acyl substitution reaction with the anti-RSV was effective, and differences in  $R_{ct}$  values from EIS data proved suitable for the system evaluation. After proper optimization, the biosensor could detect the RSV obtained through cell culture with good linearity (50.0 to 2000.0 PFU mL<sup>-1</sup>), low LOD (22.54 PFU mL<sup>-1</sup>) and high reproducibility (RSD = 9.38). The biosensor performance is outstanding since it applied an ordinary pencil carbon as an electrode, with no aid from nanoparticles, and is label-free through EIS measurements, resulting in values compared to gold standard assays.

## Declaration of Competing Interest

The authors report no conflicts of interest.

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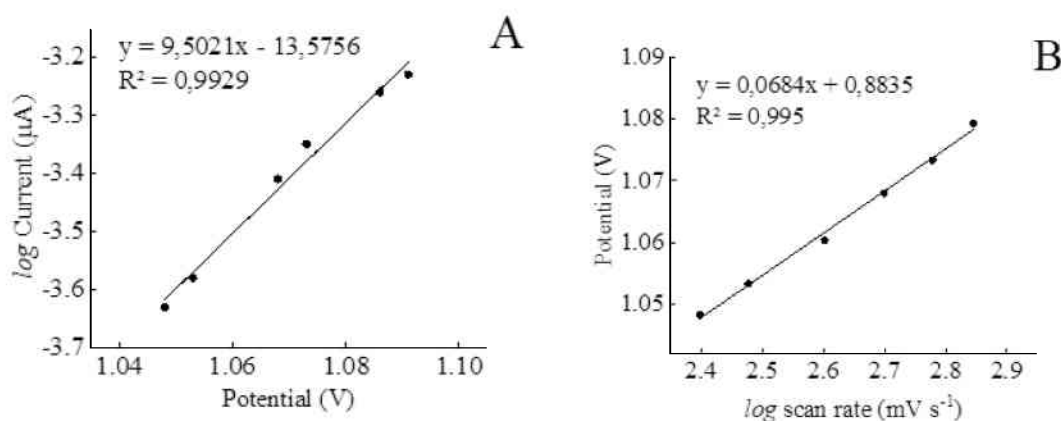


## Electronic Supplementary material

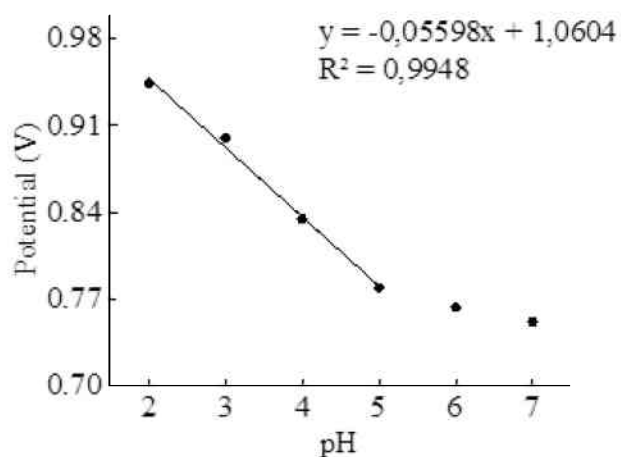
### Development of a label-free impedimetric immunosensor for the detection of respiratory syncytial virus

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Diego Leoni Franco<sup>a\*</sup>

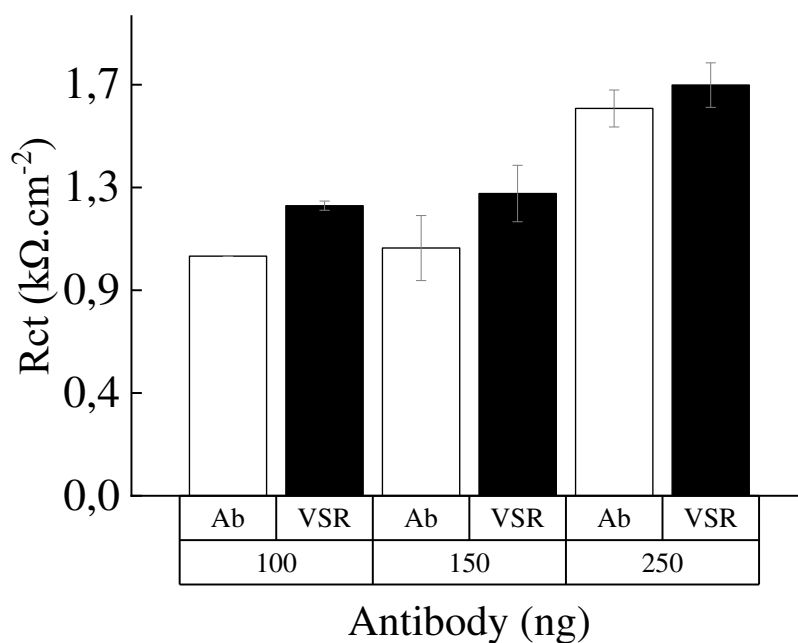
**Fig.S1.** Profile between (A) peak potential vs log of current and (B) log of scan rate vs peak potential obtained from cyclic voltammograms recorded in 3APA (2.5 mM) in sulfuric acid (0.5 M) over PGE at different scan rates. Inset: equations obtained through linear regression.



**Fig.S2.** Profile between peak potential versus pH obtained from CVs recorded in 3APA (2.5 mM) in BR buffer (0.04 M) over PCGE, 50 mV s<sup>-1</sup>. Inset: equation obtained through linear regression.



**Fig.S3.** Column chart of  $R_{ct}$  obtained through fitting using the Randles circuit from EIS data recorded in  $K_3Fe(CN)_6$  (5.0 mM) containing KCl (0.1 M). OCP at 0.25 V;  $\Delta E = 10$  mV; Frequency range: 100,000 Hz to 10 mHz. Poly-(3APA)-modified PGE with 30 cycles in a 3APA (2.5 mM) in  $H_2SO_4$  (0.5 M) solution,  $50 \text{ mV s}^{-1}$ , 100, 150, and 250 ng Ab ( $6.67 \text{ ng } \mu\text{L}^{-1}$ ) in PBS 1x pH 7.4 with an incubation time of 3 h, gly = glycine (50.0 mM) in PBS 1x pH 7.4 with an incubation time of 50 min, and VSR ( $500 \text{ PFU mL}^{-1}$ ) in PBS 1x pH 7.4 with incubation time of 3 hours.



### 3 CONCLUSÃO

Este trabalho mostrou a viabilidade da modificação de PGE com poli(3APA) e a caracterização do material formado através de CV e EIS. As respostas encontradas nos sugerem a formação de um material com a presença de grupos funcionais carboxílicos ativos para ligações covalentes. Os CVs em ácido sulfúrico antes e depois da eletropolimerização indicam que não houve resquício do monômero na superfície do PGE, além de apresentar um par redox após a modificação, indicativo da formação de um material eletroativo.

A otimização do número de varreduras demonstrou que 30 ciclos são suficientes para a modificação completa da superfície do PGE. Os estudos de pH e velocidade, juntamente com as equações de Laviron, Tafel e Nernst possibilitaram propor o mecanismo de eletropolimerização do 3APA, que inicialmente apresenta a perda de um elétron e um próton, gerando um produto com grupos carboxilas livres.

O Ab foi imobilizado de forma efetiva através de ligação covalente no PGE modificado com poli(3APA) e o vírus pode ser detectado por EIS. Foram necessárias 3 h para imobilização do anticorpo e 50 min para o bloqueio da superfície do imunossensor. A detecção de Ag pode ser feita após 30 min de imersão do imunossensor em solução de Ag.

O sistema apresentou um LOD abaixo de valores encontrados na literatura para outros biossensores eletroquímicos, demonstrando ser uma alternativa simples, barata e rápida para a detecção do RSV.

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Electroanalysis, Affinity-based electrochemical bioplatfroms, Multiplexed and/or multi-omics determinations, Clinical, Food safety

**Antonio Canals**, University of Alicante, Alacant, Spain

Sample preparation, Chromatography, Atomic spectrometry, Electroanalysis, (micro)extraction

**Doo Soo Chung**, Seoul National University, Department of Chemistry, Seoul, South Korea  
 Analytical chemistry

**Salvatore Daniele**, Ca' Foscari University of Venice, Venezia, Italy

Electroanalysis with micro and nanoelectrodes and their arrays, development and application of scanning electrochemical microscopy (SECM), development and applications of sensors in environmental, food, pharmaceutical and biological matrices, new nanomaterials for the enzyme-free detection of biological molecules, miniaturized electrochemical devices for detection of analytes in gaseous atmospheres

**Eric Deconinck**, Sciensano, Brussel, Belgium

Chromatography, (near) infrared spectroscopy, LC/GC-MS, Chemometrics, (illegal) medicines, illicit drugs

**Ludovic Duponchel**, University of Lille, Lille, France

Chemometrics / Spectroscopy

**Anastasios Economou**, National and Kapodistrian University of Athens, Athens, Greece

electrochemical sensors chemiluminescence LC-MS automated methods of analysis (flow-injection and sequential-injection analysis)

**Manuel Hernandez-Cordoba**, University of Murcia, Murcia, Spain

Gas chromatography (GC) GC/MS Liquid chromatography (LC) LC/MS Solid phase microextraction (SPME) Solvent extraction Atomic absorption spectroscopy Flame atomic absorption spectroscopy Electrothermal atomic absorption spectroscopy ICP Environmental analysis Pharmaceutical analysis Soil/sediments/plant analysis Speciation Keywords Analysis of pharmaceuticals DLLME Environmental samples Food analysis Fungicides Heavy metals Liquid microextraction Miniaturized extraction systems Pesticides Speciation Vitamins

**John Kalivas**, Idaho State University, Department of Chemistry, Pocatello, Idaho, United States of America

Chemometrics with emphasis on multivariate calibration in both methodology development and application

**Spas Kolev**, University of Melbourne School of Chemistry, Parkville, Australia

Flow analysis, Microfluidic paper-based analytical devices, Passive sampling, Membrane separation

**X. Chris Le**, University of Alberta, Edmonton, Alberta, Canada

Analytical techniques and assays for molecules of environmental and biological significance, Arsenic speciation, Exposure, Metabolism and health effects, DNA damage and protein biomarkers, Environmental health, Water and food safety

- Hian Kee Lee**, National University of Singapore, Singapore, Singapore  
Miniaturized sample preparation, Separation science, Environmental analysis
- Jin-Ming Lin**, Tsinghua University, Department of Chemistry, Beijing, China  
Microfluidic chip mass spectrometry combined with cell drug metabolism
- Baohong Liu**, Fudan University, Department of Chemistry, Shanghai, China  
Optical/electrochemical biosensing, Microfluidic analysis, Single molecule imaging, Single cell measurement
- Frank M. Matysik**, University of Regensburg Institute of Analytical Chemistry, Regensburg, Germany  
Instrumental analytical chemistry, Electroanalysis, Mass spectrometry, Hyphenated systems, Capillary electrophoresis
- Maria Minunni**, University of Florence, Department of Chemistry 'Ugo Schiff', Florence, Italy  
Chemical Sensors, surface plasmon resonance, Analytical Techniques, Aptamers, PCR, Analytical Chemistry, Lab on a Chip
- Alejandro Olivieri**, National University of Rosario, Rosario, Argentina  
Multivariate analysis, Multiway analysis, Experimental Design, Classification, Calibration.
- Sibel A. Ozkan**, Ankara University, Department of Analytical Chemistry, Ankara, Turkey  
Electroanalysis, Biosensors, Biomarkers, Nanosensors, MIP sensors, chiral sensors
- Valérie Pichon**, Sorbonne University, Paris, France & ESPCI Paris, PSL University, Paris, France  
Sample preparation, Selective extraction, Solid-phase extraction, Immunoaffinity, Aptamers, Imprinted polymers, LC/MS
- Feliciano Priego-Capote**, University of Cordoba, Department of Analytical Chemistry, Cordoba, Spain  
Metabolomics, mass spectrometry, sample preparation, automation, clinical analysis, agrofood analysis
- Marcela Segundo**, University of Porto, Faculty of Pharmacy, Porto, Portugal  
Automation, Miniaturization, Sample treatment
- Mojtaba Shamsipur**, Razi University, Kermanshah, Iran
- Kelly Smalling**, US Geological Survey New Jersey Water Science Center, Lawrenceville, New Jersey, United States of America
- Petr Solich**, Charles University, Department of Analytical Chemistry, Hradec Králové, Czechia  
Pharmaceutical analysis, Environmental analysis, Bioanalytical analysis Separation methods (LC, UHPLC), Flow methods (SIA, FIA, SIC) – development and applications Monolithic columns Drugs, Pharmaceuticals
- Robert Synovec**, University of Washington, Department of Chemistry, Seattle, Washington, United States of America  
Comprehensive two-dimensional gas chromatography with mass spectrometry, Applications in fuels, Metabolomics, Foods, Environmental, and Forensics, chemometric data analysis of chromatographic data, high-speed gas chromatography
- Beata Walczak**, University of Silesia, Katowice, Poland  
all aspects of data analysis, chemometrics
- Joseph Wang**, Arizona State University Biodesign Institute, Tempe, Arizona, United States of America  
Amperometric sensors, DNA chips, Glucose monitoring, Microfluidic devices
- Xinghua Xia**, Nanjing University, Nanjing, China  
electrochemical measurements and imaging, in situ infrared spectroelectrochemistry and nano infrared spectroelectrochemical imaging, electrochemical energy conversion and storage, micro-/ nano-fluidic devices
- Xiu-Ping Yan**, Jiangnan University School of Food Science and Technology, Wuxi, China  
Food safety, Environmental and biological analysis
- Chaoyong Yang**, Xiamen University College of Chemistry and Chemical Engineering, Xiamen, China  
aptamer, bioluminescence sensor, microfluidic and bioanalysis
- Ping Yu**, Institute of Chemistry Chinese Academy of Sciences, Beijing, China  
Electroanalysis, In vivo analysis, Ion transport, Nanopore, Biosensor
- Yong-Liang Yu**, Northeastern University, Shenyang, China  
Atomic Spectroscopy, Sample pretreatment
- Xinrong Zhang**, Tsinghua University, Department of Chemistry, Beijing, China  
Luminescence analysis, Mass spectrometry



## GUIDE FOR AUTHORS

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### INTRODUCTION

*Talanta* provides a forum for the publication of original research papers, short communications, and critical reviews in all branches of pure and applied **analytical chemistry**

#### **TYPES OF PAPERS:**

##### **Full Papers**

Papers are evaluated based on established guidelines, including the fundamental nature of the study, scientific novelty, substantial improvement or advantage over existing technology or methods, and demonstrated analytical applicability. Original research papers on fundamental studies, and on novel sensor and instrumentation developments, are encouraged. Novel or improved applications in areas such as clinical and biological chemistry, environmental analysis, geochemistry, materials science and engineering, and analytical platforms for omics development are welcome.

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Short communications comprise more abbreviated studies that demonstrate new concepts and applicability.

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Reviews including authoritative and comprehensive review articles, dedicated to a particularly important topic or field of analysis, are published regularly. In addition, shorter, concise reviews or viewpoints focusing on the current status and future prospects of a field or topic particularly relevant to the development of a new analytical methodology or to a better understanding of its fundamental underlying principles are welcome. Tutorial reviews, illustrating in depth fundamental concepts in atomic spectroscopy and analytical atomic spectroscopy, are also published.

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Analytical performance of methods should be determined, including interference and matrix effects, and methods should be **validated** by comparison with a standard method, or analysis of a certified reference material. Simple spiking recoveries may not be sufficient. The developed method should especially comprise information on selectivity, sensitivity, detection limits, accuracy, and reliability. *However, applying official validation or robustness studies to a routine method or technique does not necessarily constitute novelty.* Proper statistical treatment of the data should be provided. Relevant literature should be cited, including related publications by the authors, and authors should discuss how their proposed methodology compares with previously reported methods.

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Reference to a book:

[3] W. Strunk Jr., E.B. White, *The Elements of Style*, fourth ed., Longman, New York, 2000.

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[4] G.R. Mettam, L.B. Adams, How to prepare an electronic version of your article, in: B.S. Jones, R.Z. Smith (Eds.), *Introduction to the Electronic Age*, E-Publishing Inc., New York, 2009, pp. 281–304.

Reference to a website:

[5] Cancer Research UK, Cancer statistics reports for the UK. <http://www.cancerresearchuk.org/aboutcancer/statistics/cancerstatsreport/>, 2003 (accessed 13 March 2003).

Reference to a dataset:

[dataset] [6] M. Oguro, S. Imahiro, S. Saito, T. Nakashizuka, Mortality data for Japanese oak wilt disease and surrounding forest compositions, *Mendeley Data*, v1, 2015. <https://doi.org/10.17632/xwj98nb39r.1>.

Reference to software:

[7] E. Coon, M. Berndt, A. Jan, D. Svyatsky, A. Atchley, E. Kikinzon, D. Harp, G. Manzini, E. Shelef, K. Lipnikov, R. Garimella, C. Xu, D. Moulton, S. Karra, S. Painter, E. Jafarov, S. Molins, *Advanced Terrestrial Simulator (ATS) v0.88 (Version 0.88)*, Zenodo, March 25, 2020. <https://doi.org/10.5281/zenodo.3727209>.

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